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UNITED STATES DEPARTMENT OF AGRICULTURE
BUREAU OF PUBLIC ROADS

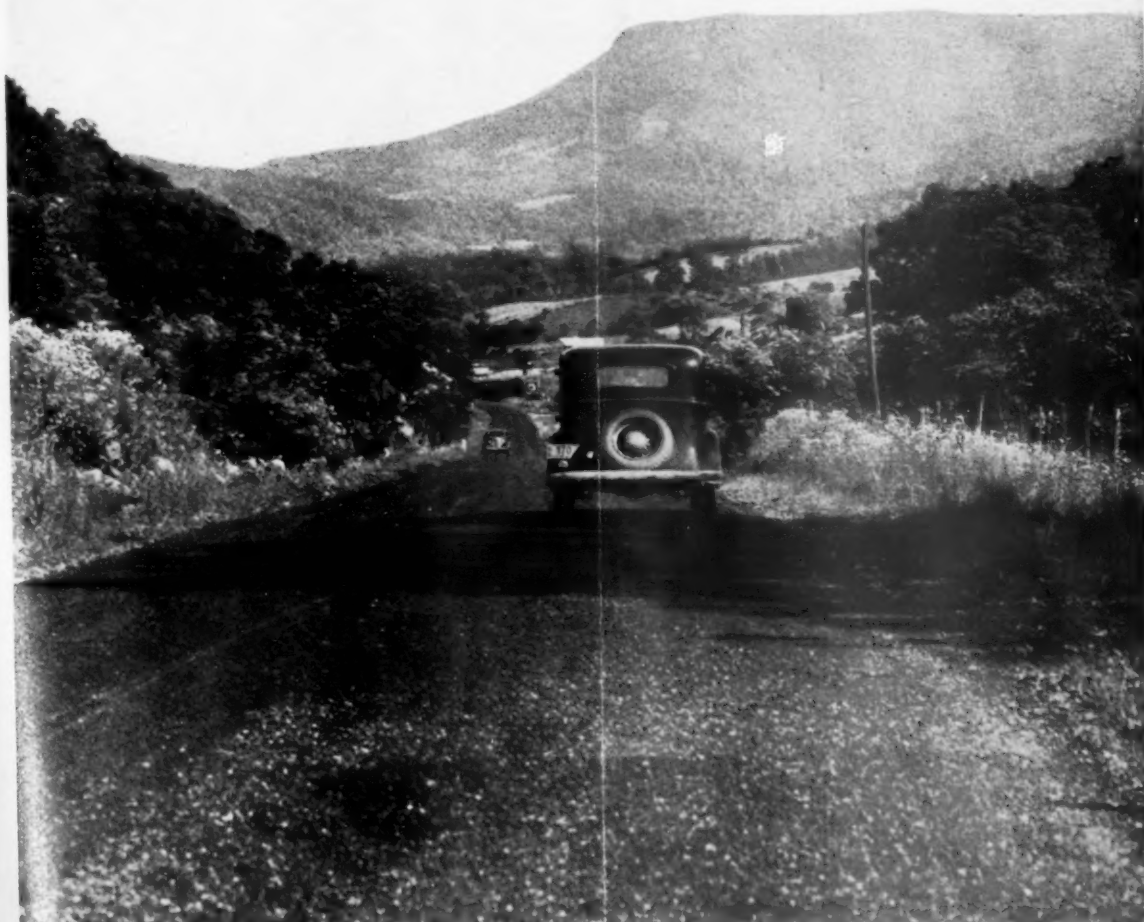


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The reports of research published in this magazine are necessarily qualified by the conditions of the tests from which the data are obtained. Whenever it is deemed possible to do so, generalizations are drawn from the results of the tests; and, unless this is done, the conclusions formulated must be considered as specifically pertinent only to described conditions.

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FURTHER STUDIES OF LIQUID ASPHALTIC ROAD MATERIALS

BY THE DIVISION OF TESTS, U. S. BUREAU OF PUBLIC ROADS

Reported by R. H. LEWIS Associate Chemist, and W. O'B. HILLMAN, Assistant Highway Engineer

IN A REPORT recently published by the Bureau of Public Roads on A Study of Some Liquid Asphaltic Materials of the Slow-Curing Type,¹ it was shown that the action of sunlight, heat, and air on these materials when exposed in relatively thin films produced residues with physical and chemical characteristics differing greatly from those of the residues developed in the usual laboratory heat tests. It was also shown that when these materials were mixed with a standard sand, molded into cylinders by the Hubbard-Field method, and subjected to the same exposure conditions as the thin films, they developed stability, or bonding strength that could not be attributed entirely to the loss of volatile matter.

MATERIALS STUDIED TYPICAL OF ALL CLASSES OF LIQUID ASPHALTIC MATERIALS FROM PRINCIPAL PRODUCING AREAS

The materials used in the earlier investigation were slow-curing liquid asphalts. They were the products of 10 refineries located in the far and middle West. In further studies conducted in 1933, that are the subject of this report, 32 materials typical of the slow-, medium-, and rapid-curing types of liquid asphaltic materials were used. These samples are identified in table 1 and were the products of 25 refineries located in all sections of the country and probably were made from petroleum of widely different bases and by various refining processes.

TABLE 1.—Products tested

Sample identification	Laboratory number	Type of material	Producer	Refinery	Location of refinery	Remarks
1.....	36626	SC-2	1	1	Oklahoma.....	
2.....	34354	SC-2	2	2	Missouri.....	
3.....	35130	SC-2	2	3	Illinois.....	
4.....	35481	SC-3	3	4	do.....	
5.....	36425	SC-3	3	5	Wyoming.....	
6.....	36045	SC-2	4	6	Arkansas.....	
7.....	36945	SC-2	5	7	do.....	
8.....	36598	SC-2	6	8	Oklahoma.....	
9.....	35195	SC-2	7	9	West Virginia.....	
10.....	35334	SC-2	8	10	Rhode Island.....	
11.....	35200	SC-1	9	11	Louisiana.....	
12.....	35396	SC-2	9	11	do.....	
13.....	36026	SC-3	9	11	do.....	
14.....	35351	SC-2	2	12	Indiana.....	
15.....	35180	SC-2	2	13	Wyoming.....	
16.....	35181	SC-2	2	14	do.....	
17.....	35103	SC-2	10	15	Indiana.....	Included in 1932 exposure.
18.....	35367	SC-2	10	16	Illinois.....	do.
19.....	34322	SC-2	11	17	do.....	do.
20.....	34059	SC-2	12	18	California.....	do.
21.....	34061	SC-2	13	19	do.....	Steam-reduced California residual oil.
22.....	34062	SC-3	13	19	do.....	
23.....	34068	SC-2	7	20	South Carolina.....	
24.....	34283	RC-2	14	21	New Jersey.....	
25.....	36762	RC-2	3	22	do.....	
26.....	35435	RC-2	2	12	Indiana.....	
27.....	35352	MC-1	2	12	do.....	
28.....	35347	RC-2	7	23	Maryland.....	
29.....	36071	RC-2	15	24	do.....	
30.....	34907	MC-2	16	25	Wyoming.....	
31.....	35916	MC-2	2	13	do.....	
32.....	35922	MC-2	13	19	California.....	

¹ Furol viscosity was below specification limit.

² Furol viscosity was above specification limit.

³ Penetration of distillation residue was below specification limit.

⁴ R. H. Lewis and W. O'B. Hillman, PUBLIC ROADS, June 1934, vol. 15, no. 4.

As shown in table 1, 23 of the materials were of the slow-curing type, of which 4 samples were tested in 1932 and were included in this study for comparative purposes. Four were of the medium-curing type and five were rapid-curing products. They met the provisional specifications, as given in table 2, of the Bureau of Public Roads and the Asphalt Institute, except as noted in table 1. Samples 21 and 22 were the only slow-curing products for which there was any definite information as to origin or method of manufacture. Both of these materials were steam-reduced California residuals without subsequent blending. All of the rapid-curing products were prepared from 85-100 penetration asphalt and solvent naphtha. The composition of sample 27 was unknown but the other medium-curing products, samples 30, 31, and 32, were, respectively, 110-120 penetration asphalt, 94+ asphaltic road oil, and 100-120 penetration asphalt fluxed with a heavy grade of kerosene. These three medium-curing materials, although subjected to all of the laboratory tests, were exposed only under special conditions that are described later in this report.

TABLE 2.—Specification requirements for grades of liquid asphaltic road materials investigated

	SC-1	SC-2	SC-3	MC-1	MC-2	RC-2
Flash point, °F.....	150+	200+	200+	—	150+	80+
Furol viscosity at 77° F., seconds.....	20-150	—	—	40-150	—	—
Furol viscosity at 122° F., seconds.....	—	200-320	—	—	—	200-400
Furol viscosity at 140° F., seconds.....	—	—	150-300	—	150-250	—
Total distillate to 437° F., percent by volume.....	—	2-	2-	10-	2-	10+
Total distillate to 600° F., percent by volume.....	—	15-	10-	25+	10-20	20+
Total distillate to 680° F., percent by volume.....	50-	25-	20-	50-	27-	35-
Tests on distillation residue:						
Float at 122° F., seconds.....	50-	25+	25+	—	—	—
Penetration at 77° F.....	—	—	—	70-300	100-300	60-120
Ductility at 77° F., centimeters.....	—	—	—	60+	60+	60+
Soluble in CS ₂ , percent.....	99.0+	99.0+	99.0+	99.5+	99.5+	99.5+

The test procedure followed that of the 1932 study except that the fixed-carbon test was omitted as the changes in inherent characteristics that occurred under laboratory and exposure conditions appeared to be more strikingly illustrated by the test for solubility in 86° B. naphtha. Two new tests were added. The Oliensis test² for heterogeneity was made on the original materials and on all of the residues, except those from the 50-gram oven-loss test and the 10-week exposure test, and the original materials were examined microscopically. The results of the tests on the original materials are given in table 3 and a detailed analysis of the residues obtained in the routine laboratory tests is given in table 4.

² A qualitative test for determining the degree of heterogeneity of asphalts. G. L. Oliensis, Proc. A. S. T. M., vol. 33, pp. 715-728.

TABLE 3.—Results of tests on original materials

Sample identification	Specific gravity at 77° F.		Flash point		Furoil viscosity		Organic matter insoluble in CS ₂		Organic matter insoluble in CCl ₄		Insoluble in 80° B. naphtha		Characteristic by Olenis test ¹		Microscopic smear test		Distillation										Loss at 325° F., 5 hours						Asphaltic residue																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																									
																	Initial boiling point		Distillate by volume		Distillate to 680° F. (by weight)		Loss on cooling (by weight)		Total loss (by weight)		Tests on distillate		Penetration of residue at 77° F., 100 g., 5 sec.		50-gram sample				20-gram sample																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																							
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¹H = Heterogeneous; O = Homogeneous; SH = Slightly heterogeneous.

DISTILLATION CURVES CLEARLY DISTINGUISH SLOW-, MEDIUM-, AND RAPID-CURING MATERIALS

The distillation curves for the various materials, plotted in figure 1, serve to distinguish the three classes of materials according to their curing properties. It will be noted that the initial boiling points of the slow-curing products, samples 1 to 23, inclusive, were all above 450° F. and in only two instances were they less than 500° F. The medium-curing materials, samples 27, 30, 31, and 32, had initial boiling points as low as 360° F. with none above 450° F.; and the rapid-curing products, samples 24, 25, 26, 28, and 29, all had initial boiling points below 370° F.

With all three classes of materials, the percentage of distillate increased as the temperature increased but not at the same rate. For the slow-curing materials, the rate continued fairly uniform up to 680° F., but with the medium- and rapid-curing products it decreased as the temperature approached 680° F. The decrease in rate was more pronounced for the rapid-curing products. The rate of distillation may be illustrated by expressing the amount of distillate off at any temperature in terms of the total distillate recovered at 680° F. For the medium-curing materials the amount at 600° F. was 65 to 85 percent of the total. For rapid-curing materials the amount was 0 to 50 percent at 374° F., 41 to 85 percent at 437° F., and 88 to 94 percent at 600° F. The residues from the distillation of the slow-curing products were all fluid, while those of the medium- and rapid-curing materials were semisolid.

That portion of the total loss in the distillation test designated as loss on cooling depends upon the amount of material boiling off immediately above 680° F. As would be expected from the slope of the distillation curves, this loss was greater for the slow-curing products than for the medium- and rapid-curing products. It ranged from 1.2 percent to 5.8 percent with an average of 4.5 percent for the slow-curing products, from 2.9 percent to 4.2 percent with an average of 3.4 percent for the medium-curing products, and from 1.5 percent to 3.1 percent with an average of 2.2 percent for the rapid-curing products.

The total volatile matter in the slow-curing materials as determined by the distillation test, including both the distillate recovered and the loss on cooling, ranged from 2.8 percent to 25 percent with an average of 13.8 percent for the group. The average loss on cooling of 4.5 percent actually represented 36 percent of the total volatile matter in the average slow-curing material used in this study. While this loss on cooling may be unimportant in estimating the relative volatility of various liquid asphalts, it must be considered if the results of the distillation test are to be compared directly with the results of other laboratory and exposure tests.

The different classes of materials are also readily identified by the results of the volatilization and asphaltic-residue tests. The slow-curing products lost in the 50-gram volatilization test from 53 to 76 percent, the medium-curing products from 73 to 80 percent, and the rapid-curing products from 92 to 97

TABLE 4.—Results of tests on laboratory residues

Sample identification	Distillation										Loss at 325° F., 5 hours										Asphaltic residue									
	Tests on residue										50-gram sample										20-gram sample									
											Tests on residue										Tests on residue									
	Total loss in distillation test	Penetration	Ductility at 5 centimeters per minute	Organic matter insoluble in CS ₂	Organic matter insoluble in CCl ₄	Insoluble in 86° B. naphtha	Loss	Penetration	Ductility at 5 centimeters per minute	Organic matter insoluble in CS ₂	Organic matter insoluble in CCl ₄	Insoluble in 86° B. naphtha	Loss	Penetration	Ductility at 5 centimeters per minute	Organic matter insoluble in CS ₂	Organic matter insoluble in CCl ₄	Insoluble in 86° B. naphtha	Loss	Penetration	Ductility at 5 centimeters per minute	Organic matter insoluble in CS ₂	Organic matter insoluble in CCl ₄	Insoluble in 86° B. naphtha						
	Pct. Sec.	At 122° F.	At 77° F., 100 g., 5 sec.	At 32° F., 200 g., 60 sec.	Softening point	At 77° F.	At 34°-35° F.	At 77° F.	At 34°-35° F.	At 77° F.	At 34°-35° F.	At 77° F.	At 34°-35° F.	At 77° F.	At 34°-35° F.	At 77° F.	At 34°-35° F.	At 77° F.	At 34°-35° F.	At 77° F.	At 34°-35° F.	At 77° F.	At 34°-35° F.	At 77° F.	At 34°-35° F.					
1	2.8	24																												
2	11.0	60																												
3	6.8	43																												
4	5.7	45																												
5	13.6	58																												
6	7.4	21																												
7	23.4	62																												
8	6.4	27																												
9	10.9	44																												
10	22.3	87																												
11	25.0	34																												
12	17.9	38																												
13	9.3	37																												
14	5.1	36																												
15	15.4	41																												
16	6.8	25																												
17	20.6	110																												
18	17.7	60																												
19	9.9	57																												
20	13.9	50																												
21	14.8	41																												
22	9.8	56																												
23	11.5	27																												
24	24.8	76	18	117	110+	0.0																								
25	23.8	77	28	122	110+	4.5																								
26	23.4	84	28	116	110+	4.0																								
27	30.1	87	15	109	110+	0																								
28	21.1	65	20	122	110+	0																								
29	23.5	72	25	121	110+	4.5																								
30	25.6	81	25	117	99	2.5																								
31	20.2	205	41	101	110+	15.0																								
32	22.1	298	55	93	75	90+																								

percent as much as they lost in the 20-gram volatilization test. The residues from both volatilization tests of the slow-curing products were fluid; those of the medium-curing products were fluid in the 50-gram volatilization test and semisolid in the 20-gram volatilization test; and those of the rapid-curing products were both semisolid. For the slow-curing products, the loss in neither volatilization test amounted to as much as that in the asphaltic residue test. For the medium-curing products, the loss in the 20-gram volatilization test approximated the loss in the asphaltic-residue test, while for the rapid-curing products the losses in all tests, including the distillation test, were approximately the same. The loss in the 20-gram volatilization test, while always greater than that in the 50-gram volatilization test, was always less than in the distillation test. The residue from the 20-gram volatilization test may, however, be harder or softer than the residue from distillation.

The slow-curing products were reduced to a residue of 100 penetration in from 30 to 420 minutes, with an average of 126 minutes. In producing residues of the same consistency medium-curing products took from 16 to 30 minutes, with an average of 23 minutes and the rapid-curing products took from 11 to 15 minutes, with an average of 13 minutes. When making the asphaltic-residue test on the rapid-curing products, difficulty was experienced in obtaining a residue that was

not too hard, since the high volatility of the solvent caused the cut-back materials to be reduced to 100 penetration before the temperature for making the test was reached.

MAJORITY OF MATERIALS SHOWED A CLEAR FIELD ON MICROSCOPIC EXAMINATION

The microscopic test, adopted by one State highway department for the detection of cracking-coil products, was made on all materials. The specification of the State did not set up an exact procedure. It merely stated that a freshly prepared smear of the asphaltic material diluted with carbon tetrachloride should show a clear field free from carbonaceous matter when subjected to a magnification of 200 diameters. In the work covered by this report, the test was standardized by using 2 parts by weight of carbon tetrachloride and 1 part by weight of asphaltic material in preparing the slides for observation. When prepared in this manner all of the materials but seven showed a clear field. In preparing the slides for the photomicrographs illustrated in figure 2, 6 parts by weight of carbon tetrachloride and 1 of the asphaltic material were used.

Because carbon tetrachloride has both solvent and flocculent properties, its use as a diluent was questioned. Therefore, slides were also prepared with the undiluted materials and it was found that only those materials that showed flecks when undiluted showed

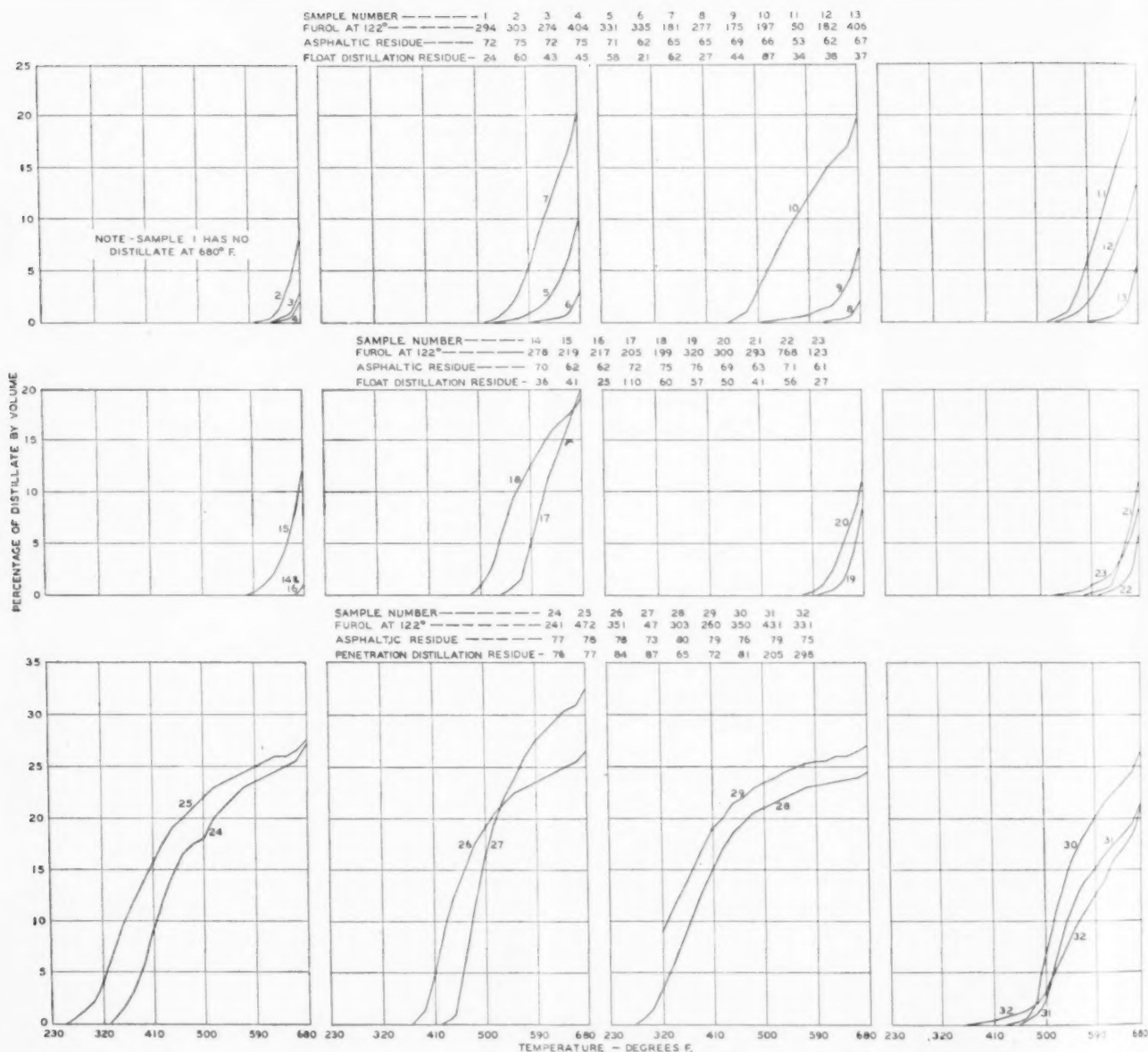


FIGURE 1.—RELATION OF PERCENTAGE OF DISTILLATE BY VOLUME TO DISTILLING TEMPERATURE.

them in the 2 to 1 and 6 to 1 dilutions. This indicates that the insoluble matter was already flocculated and that carbon tetrachloride, in the quantities used, did not precipitate carbonaceous flecks in those materials not containing them when undiluted. Recently a sample was tested that contained flecks when undiluted that disappeared on dilution with carbon tetrachloride; and another sample that was practically clear when undiluted contained flecks when diluted. In the first case, the poor solvent properties of the distillate used in the material were responsible for incomplete solution of the base asphalt, that immediately went into solution with the addition of carbon tetrachloride. In the second case it was quite evident that the carbon tetrachloride acted as a flocculent. However, since 6 of the 7 samples that contained carbonaceous material had relatively high percentages of material insoluble in carbon tetrachloride, it is

probable that the flecks shown are particles of free carbon and carbones.

In conducting the exposure tests three samples of each material were placed in seamless, flat-bottom tins having a diameter of 5½ inches and a depth of five-eighths inch. Fifty cubic centimeters of material were used to obtain a uniform film or layer thickness of about one-eighth inch. The samples were then placed in exposure boxes made of wood. A plate-glass cover resting on strips of felt fastened to the edges of each box made a tight joint and excluded all dust and dirt. A current of air was passed through a wash bottle containing sulphuric acid to remove dust and eliminate moisture, and was admitted through the bottom of the boxes and escaped through slots in the sides, thus serving to carry off the vapors formed. The slots were protected from rain by thin boards extending from the top of the box downward at an

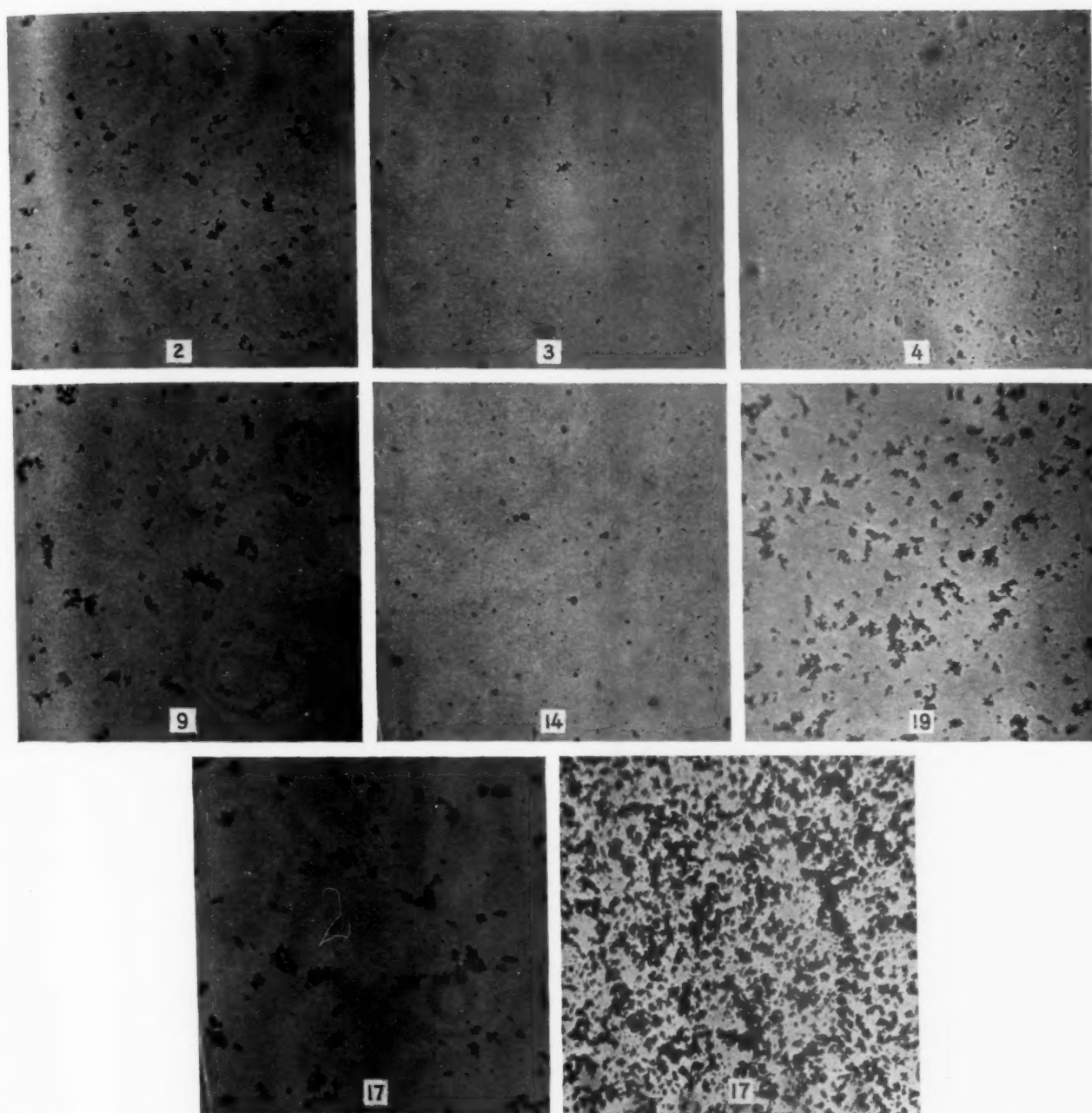


FIGURE 2.—PHOTOMICROGRAPHS OF MATERIALS CONTAINING CARBONACEOUS MATTER (MAGNIFIED 200 TIMES). THE LOWER TWO ILLUSTRATIONS SHOW RESULTS OBTAINED WITH SAMPLE 17; IN THE ONE ON THE LEFT THE MATERIAL WAS DILUTED WITH CARBON TETRACHLORIDE AND THE ONE ON THE RIGHT SHOWS UNDILUTED MATERIAL.

angle of about 45° . Cotton batting inserted in the slots excluded dust. A thermometer in each box provided a means of determining the temperature. The assembly of the boxes is shown in figure 3.

DIFFERENCES FOUND BETWEEN SLOW-CURING AND CUT-BACK PRODUCTS AFTER EXPOSURE

The samples were placed in the boxes on June 15, 1933, and were weighed periodically to determine the loss in weight. A complete set of samples was removed and tested at the end of 5 weeks, another set at 10 weeks, and the last set at 15 weeks. The temperature of the boxes was dependent entirely upon the radiant

heat of the sun and varied with the amount of sunshine. On clear days the temperature was extremely high, the maximum recorded being 196°F. , but on days with no sunshine the temperature in the boxes was the same as that of the air. During the period of exposure the average maximum daily air temperature was 85°F. The samples exposed for 5 weeks were subjected to 333 hours of sunlight and those exposed for 10 and 15 weeks were subjected to 611 and 866 hours of sunlight, respectively. The percentage of loss at different periods of exposure is given in table 5 and the results of tests on the residues are given in table 6. Photographs of typical surfaces at the end of 15 weeks are shown in figure 4.

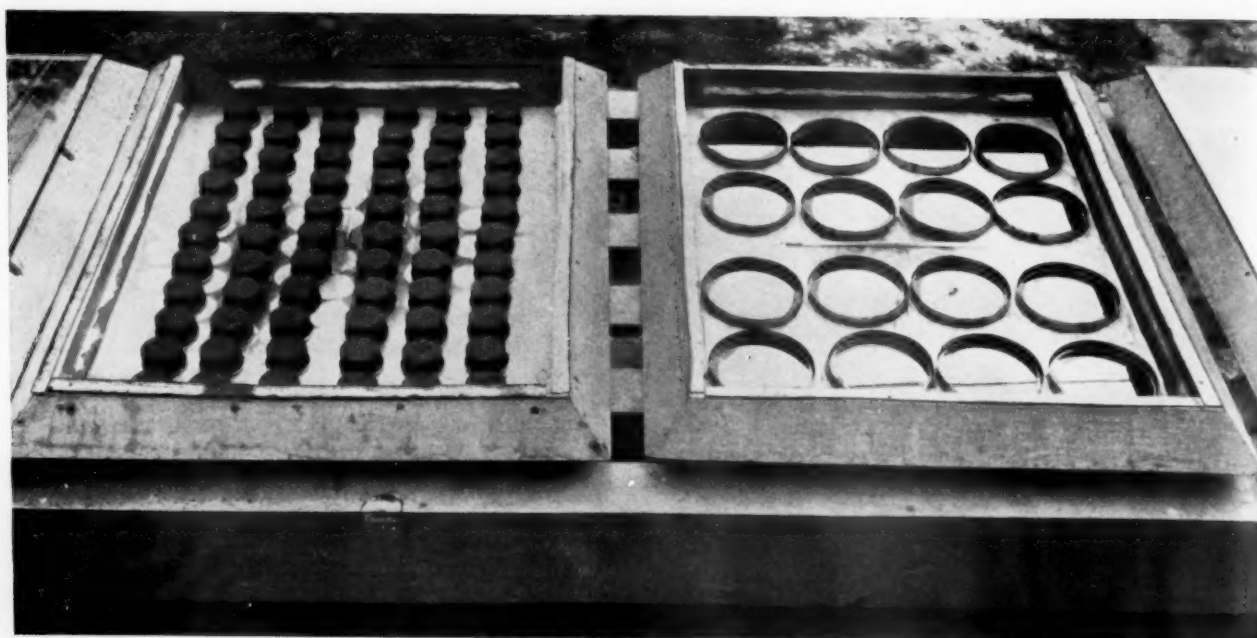


FIGURE 3.—STABILITY SPECIMENS AND THIN FILMS OF MATERIAL EXPOSED TO SUNLIGHT.

While the majority of the materials progressively lost weight during the period of exposure, a number of them actually gained at first, although they later lost more than the amounts gained. An exception to this was sample 1, which had gained 3.6 percent at the end of 8 days and at the end of 15 weeks still showed a slight gain. The samples exposed for 15 weeks were used in determining the loss at 2, 8, 15, 22, 50, and 105 days, while the percentage of loss given for the 35- and 70-day exposures was based on samples used for test at the end of 5 and 10 weeks, respectively. This was done to eliminate errors in calculating the results of subsequent tests made upon the respective residues and accounts for slight variations that may appear to indicate gains instead of losses.

As was expected, the cut-back products lost weight very rapidly. At the end of 2 days they had lost from 86 to 92 percent of their maximum loss with an average of 89 percent, and at the end of 35 days they had lost from 97 to 100 percent with an average of 99 percent. For the slow-curing products the rate of loss was much less, but was considerably more variable. In 2 days those samples that had undergone a loss had lost from 3 to 60 percent of their maximum loss with an average of 35 percent. In 15 days they had lost from 16 to 84 percent with an average of 50 percent, in 35 days from 63 to 100 percent with an average of 82 percent, and in 70 days from 74 to 100 percent with an average of 89 percent. Some idea of the relative speed of curing or volatility of the two types of material may be obtained by comparing their losses under these exposure conditions. The slow-curing materials took 70 days to lose an average of 89 percent of their volatile matter, while the cut-back materials underwent the same percentage loss in 2 days.

LOSS IN DISTILLATION TEST MOST NEARLY APPROXIMATED LOSS IN 15 WEEKS' EXPOSURE FOR ALL TYPES OF MATERIALS

Figure 5 shows the relation between the percentage of loss upon exposure and loss in the distillation test,

TABLE 5.—Loss in thin film exposure

Sample identification	Loss on exposure for—							
	2 days	8 days	15 days	22 days	35 days	50 days	70 days	105 days
	Percent	Percent	Percent	Percent	Percent	Percent	Percent	Percent
1.....	-2.6	-3.6	-3.4	-3.4	-3.0	-2.7	-2.7	-0.9
2.....	1.9	4.3	5.7	7.6	9.5	10.3	10.2	12.4
3.....	-2	2.0	2.9	4.7	6.6	8.0	8.0	9.4
4.....	1.5	1.7	3.2	5.1	4.4	5.8	5.8
5.....	2.6	5.4	6.5	8.2	9.5	9.8	9.9	11.5
6.....	-3	1.2	2.0	4.1	5.1	6.1	8.5	8.2
7.....	7.8	12.9	14.6	17.7	20.2	20.0	21.0	21.5
8.....	-1.8	2.3	1.8	2.0
9.....	.3	2.4	4.0	5.3	8.1	8.2	9.6	10.7
10.....	11.0	14.4	14.5	15.7	16.4	16.7	16.8	17.3
11.....	7.8	14.7	16.4	18.2	20.4	21.0	21.6	22.3
12.....	4.6	8.9	9.4	11.8	13.7	13.4	13.8	15.2
13.....	.4	2.6	4.3	6.1	7.3	10.1	8.5	9.8
14.....	-1.3	.2	1.3	3.4	5.2	6.9	6.0	8.1
15.....	1.9	5.5	6.9	8.4	9.3	12.1	10.6	13.3
16.....	-3.5	-1.5	-1.0	1.8	4.6	5.7	5.7	6.7
17.....	8.6	12.6	14.4	15.9	18.5	18.9	18.7	19.5
18.....	11.5	15.7	16.5	17.4	18.2	19.2	17.7	19.0
19.....	1.4	5.0	6.1	8.6	10.1	12.4	12.0	13.8
20.....	2.3	6.8	7.8	9.8	11.7	12.6	11.5	13.6
21.....	-1.1	3.7	6.4	8.2	12.5	13.6	12.9	14.3
22.....	.6	3.4	5.1	6.7	8.5	9.8	9.1	10.5
23.....	1.3	3.7	4.6	6.6	8.3	10.6	13.1	11.2
24.....	19.4	21.2	21.5	22.0	22.2	22.2	22.5	22.5
25.....	20.1	21.6	21.9	22.1	22.2	22.2	22.0	22.2
26.....	20.5	21.6	21.8	22.1	22.4	22.1	22.1	22.0
27.....	26.1	27.1	27.3	27.5	27.6	27.6	28.3	27.8
28.....	17.7	19.6	19.9	20.0	19.9	20.1	20.0	20.1
29.....	19.2	20.8	21.3	21.4	21.6	21.4	21.4	21.4

loss in the two oven tests, and loss in the asphaltic-residue test. Figure 5 indicates that for the slow-curing products the loss in 15 weeks' exposure was about $2\frac{1}{2}$ times as great as that in the oven test on a 50-gram sample, about $1\frac{1}{2}$ times as great as that in the oven test on a 20-gram sample, and about the same as the loss in the distillation test. No relationship was apparent between the loss in the exposure and asphaltic-residue tests. The loss in the latter test was, however, invariably greater, ranging from $1\frac{1}{3}$ to 14 times the loss occurring in 15 weeks' exposure with an average of $2\frac{1}{2}$ times this loss. For the cut-back products the loss in all tests was approximately the same. In all the materials studied, both in 1932 and 1933, the total loss in

TABLE 6.—Results of tests on exposure residues

Sample identification	5-week exposure														10-week exposure														15-week exposure																	
	Loss		Penetration		Ductility at 5 centimeters per minute		Organic matter insoluble in CS ₂		Organic matter insoluble in CCl ₄		Insoluble in 86° B. naphtha		Loss		Penetration		Ductility at 5 centimeters per minute		Organic matter insoluble in CS ₂		Organic matter insoluble in CCl ₄		Insoluble in 86° B. naphtha		Loss		Penetration		Ductility at 5 centimeters per minute		Organic matter insoluble in CS ₂		Organic matter insoluble in CCl ₄		Insoluble in 86° B. naphtha											
	Float at 122° F.		At 77° F., 100 g., 5 sec.		At 32° F., 200 g., 60 sec.		Softening point		At 77° F.		At 34°-35° F.		At 77° F., 100 g., 5 sec.		At 32° F., 200 g., 60 sec.		Softening point		At 77° F.		At 34°-35° F.		Organic matter insoluble in CS ₂		Organic matter insoluble in CCl ₄		Organic matter insoluble in 86° B. naphtha		At 77° F., 100 g., 5 sec.		At 32° F., 200 g., 60 sec.		Softening point		At 77° F.		At 34°-35° F.		Organic matter insoluble in CS ₂		Organic matter insoluble in CCl ₄		Organic matter insoluble in 86° B. naphtha			
	Pct.	Sec.	°F.	Cm	Cm	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	°F.	Cm	Cm	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	°F.	Cm	Cm	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.	Pct.			
1	2.6	68																																												
2	9.5		86	34	108	63.0	1.5	2.74	9.23	30.2	10.2	33	15	142	0.5	0.0	5.08	13.34	34.5	12.4	20	9	154	3	0	5.71	15.36	40.6																		
3	6.6	153			95			2.93	8.44	28.0	8.0	62	22	127	5.0	0	4.46	11.29	31.3	9.4	49	15	144	8	0	5.69	12.08	36.6																		
4	5.1		152	39	106	22.0	4.0	1.28	5.95	25.6	5.8	112	33	134	4.3	0	2.46	7.41	29.3	5.8	59	18	153	5	0	3.36	8.40	34.6																		
5	9.5	200			96					25.9	9.9	97	37	118	87.0	3.8			30.5	11.5	52	23	135	21.0	5	0	3.36	8.40	34.6																	
6	-0.3	34								15.3	7.4								17.0	8.2																										
7	20.2	123								21.0	21.0	163							22.1	21.5	195	76	107	85.0	6.5	0	1.48	25.0																		
8	2.3	163								25.9	9.9	112	33	134	4.3	0	2.46	7.41	29.3	5.8	59	18	153	5	0	3.36	8.40	34.6																		
9	8.1	1,000+								21.0	21.0	163							22.1	21.5	195	76	107	85.0	6.5	0	1.48	25.0																		
10	16.4	189	80	106	95.0	5.0		3.22	7.51	26.4	9.6	1,000+	300+						29.4	17.3	199	60	156	5	0	5.48	11.34	32.5																		
11	20.4	131								26.6	16.8		95	53	129	10.0	3.5		29.4	17.3	199	60	156	5	0	5.48	11.34	32.5																		
12	13.7	110								23.8	21.6		73	35	121	17.0	1.5		29.3	22.3	199	60	156	5	0	5.48	11.34	32.5																		
13	7.3	129								21.0	13.8	203							23.9	15.2	175	84	115	41.0	4.5	0	1.48	25.0																		
14	5.2	105						1.63	5.69	29.4	6.0		126	31	103	105.0	2.5	3.08	7.94	31.5	8.1	80	21	112	97.0	3.07	8.61	33.4																		
15	9.3	150								25.9	10.6		105	47	117	53.0	3.5		30.3	13.3	64	30	133	8.0	5	0	2.33	33.2																		
16	4.6	70								19.9	5.7	218							23.7	6.7	245	99	109	47.0	5.0	0	0.88	25.1																		
17	18.5	51	22	136	6.0	0	1.81	7.57		30.8	18.7		36	14	170	5	0	3.17	9.66	36.2	19.5	29	12	168	5	0	3.30	9.88	35.4																	
18	18.2	197								14.8	17.7		152	43	105	110+	7.5		17.3	19.0	93	28	116	110+	4.3	0	1.48	25.0																		
19	10.1	156	28	98		1.0	4.60	11.69		34.4	12.0		44	10	121	110+	0	6.50	16.02	38.7	13.8	29	7	128	110+	0	7.50	18.15	39.9																	
20	11.7	179								21.4	11.5		136	41	112	90.0	3.8		24.2	13.6	84	28	120	72.0	3.0	0	0.99	27.2																		
21	12.5	166								18.2	12.9	170							22.3	14.3	166	48	103	110+	8.5	0	0.88	25.1																		
22	8.5	150								19.1	9.1		157	42	104	110+	7.3		23.2	10.5	112	28	111	110+	5.0	0	1.48	25.0																		
23	8.3	116								15	24.7	13.1	27	14	147	1.0	0	2.91	32.8	11.2	49	22	129	7.0	0	0.28	8.1	30.2																		
24	22.2	21	9	146	17.0	0	21	28	32.4	22.5			12	10	161	5.5	0	4.45	36.9	22.5	13	6	100	5.0	0	0.63	7.3	37.4																		
25	22.2	29	14	155	5.0	0	38	50	31.7	22.0			25	13	158	4.8	0	4.43	33.9	22.2	24	13	163	4.0	0	0.72	7.9	34.2																		
26	22.4	26	14	151	4.5	0	23	30	29.0	22.1			20	13	163	3.5	0	3.5	31.5	22.0	22	13	161	3.8	0	0.47	5.1	31.4																		
27	27.6	18	7	138	4.0	0	32	2.21	27.9	28.3			12	6	158	0	0	34	2.26	30.5	27.8	14	5	153	3	0	0.57	2.40	31.6																	
28	19.9	21	9	151	7.0	0	23	23	28.2	20.0			16	9	163	4.5	0	4.45	31.0	20.1	17	8	168	3.8	0	0.44	5.2	32.8																		
29	21.6	23	13	154	7.0	0	39	39	34.7	21.4			18	13	169	4.5	0	5.1	50	27.9	21.4	17	11	172	4.0	0	0.69	8.3	39.3																	

¹ The 10- and 15-week exposure residues of this sample were nonhomogeneous.

² None of the residues after exposure of this sample was homogeneous.

the distillation test most nearly approximated the loss in 15 weeks' exposure.

At the end of 15 weeks' exposure the surfaces of the samples varied greatly in appearance, as shown by the typical photographs in figure 4. The appearance of sample 14 was typical of samples 1, 2, 3, 4, 5, 6, 7, 9, 12, 18, 19, 20, 21, and 22. However, there were some variations in actual appearance that could not be shown in a photograph. Samples 1, 2, and 3 had mottled, slightly greasy surfaces. Samples 4, 9, and 19 had mottled, slightly greasy surfaces that were slightly iridescent. Samples 14, 21, and 22 had mottled, iridescent surfaces that were not greasy. Sample 5 had a uniformly mottled appearance neither iridescent nor greasy. Samples 6, 7, 12, 18, and 20 were smooth and glossy.

Sample 23 was mottled and slightly greasy, like samples 1, 2, and 3, but had some dull areas as shown. Sample 13, while smooth and glossy, had some dull areas as shown; sample 16 was similar although it was slightly mottled.

Samples 10, 11, and 15 were checked over most of the area of their surfaces; their condition is shown by the photograph of sample 15. The unchecked areas in samples 10 and 11 were glossy but in sample 15 the surface was neither dull nor glossy. Samples 8 and 17 had very rough surfaces, shrunken and pitted as shown by their photographs. The material in the bottom of the cracks was soft but the outer surface was very hard. The cut-back products were all very rough and wrinkled, as shown by the photographs of samples 26, 27, and 28.

The material in the cracks was glossy while the outer surface was dull.

At the end of 15 weeks all of the cut-back products were exceedingly hard. As they were originally combinations of semisolid asphalt and volatile fluxes and became semisolid in the laboratory tests, it would be expected that a semisolid residue would rapidly develop upon exposure. It has been believed generally that the asphalt base used in cut-back materials is relatively resistant to changes caused by weathering and that after the cutting medium is removed the character of the material changes but little. The six cut-back materials used in this study, while losing but little weight after 2 days, were much harder and had much less ductility at the end of 5 weeks' exposure than the asphalt used as base material.

The slow-curing products varied as greatly in consistency as in the rate and amount of volatile matter given off during exposure. At the end of 5 weeks all of them were harder than the residues from the distillation and oven tests, but only 5 had a penetration at 77° F. of less than 200. The consistency varied from a float of 34 seconds at 122° F. to a penetration of 51 at 77° F. At the end of 15 weeks only 3 materials had a penetration at 77° F. of over 200 and one of these materials had disintegrated to such an extent that no true penetration test was possible. The consistency ranged from a float of 71 seconds at 122° F. to a penetration of 20 at 77° F. In the previous work, the slow-curing materials all developed residues after 15

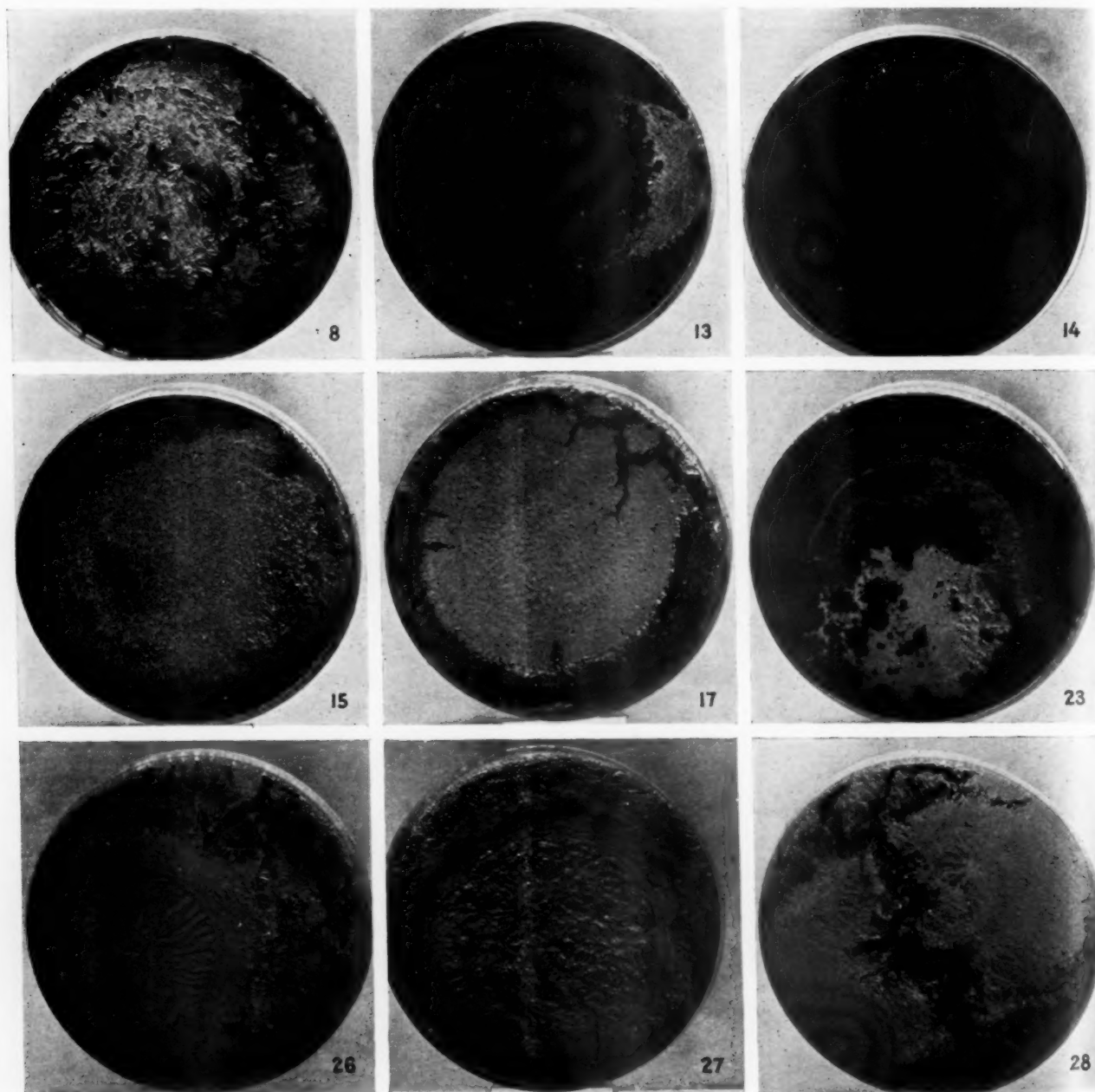


FIGURE 4.—CONDITION OF SURFACES AFTER 15 WEEKS OF EXPOSURE.

weeks that were as hard as or harder than their asphaltic residues of 100 penetration. It seems that it is impossible to predict, from the results of any of the laboratory tests, the consistency of the residues after exposure. Generally, however, those materials that took a long time to be reduced to 100 penetration in the asphaltic-residue test and those whose residues from the distillation and oven tests had a low float-test value were the softest or most fluid at the end of 15 weeks.

ASPHALTIC RESIDUES DIFFERED IN CERTAIN RESPECTS FROM RESIDUES AFTER EXPOSURE OF ABOUT THE SAME PENETRATION

A comparison of the residues after exposure and asphaltic residues is of interest. All but 5 of the asphaltic residues had ductilities at 77° F. of more than 110 cen-

timeters and 15 had ductilities at 34°–35° F. of 3½ centimeters or more. After 15 weeks' exposure only 7 products had ductilities at 77° F. over 50 and only 4 over 110. Only 6 had ductilities at 34°–35° F. over 3½. These differences in ductilities may, in a number of instances, have been caused by differences in the consistencies of the residues after exposure and asphaltic residues.

However, in some cases one of the residues from exposure had approximately the same penetration as the asphaltic residue and the laboratory residues and residues after exposure may be compared directly. Table 7 shows the results of tests on both such residues and indicates that the two residues from the same material, although having the same consistency, varied considerably in other respects.

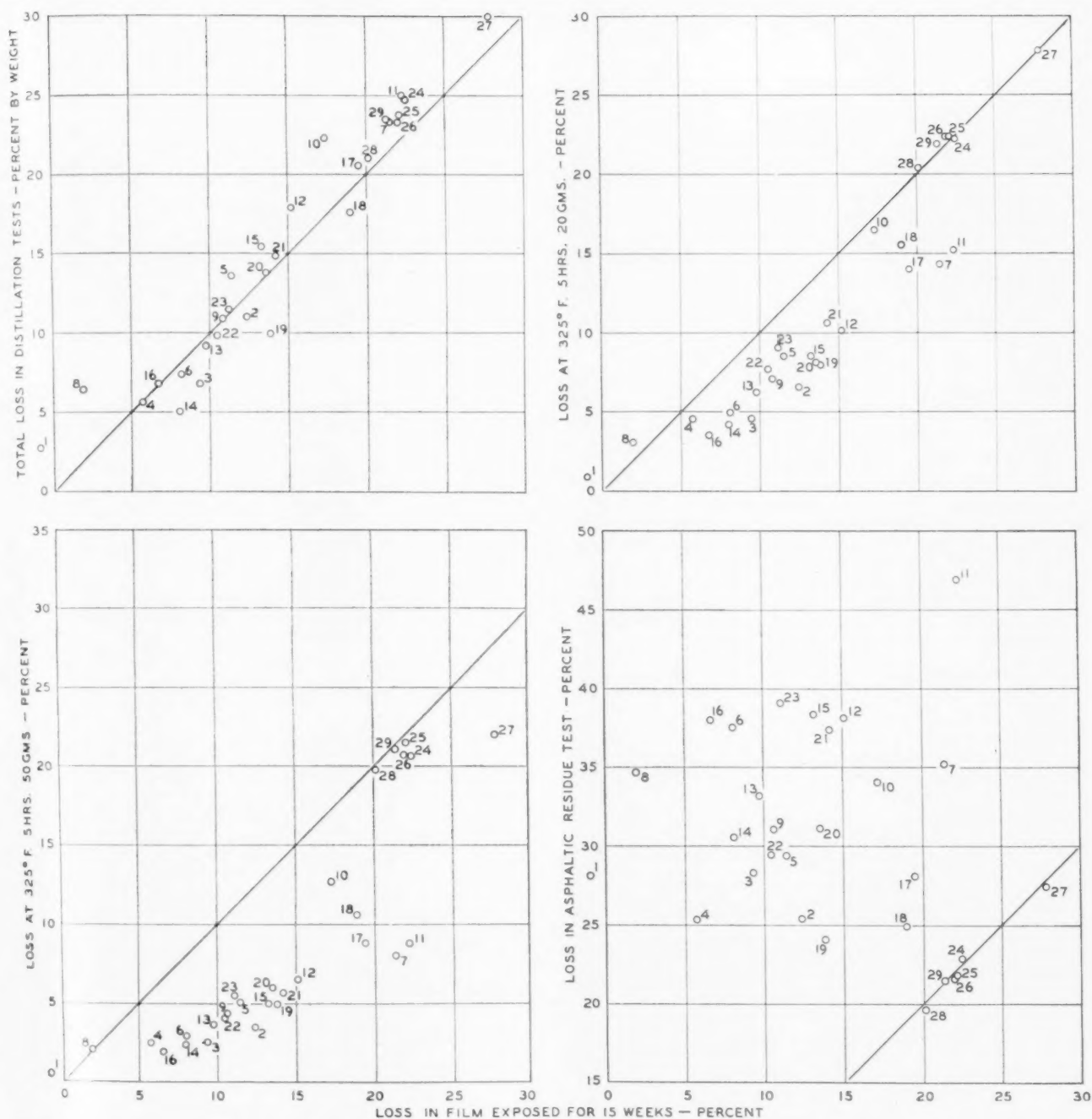


FIGURE 5.—COMPARISON OF THE PERCENTAGE OF LOSS AFTER 15 WEEKS OF EXPOSURE WITH LOSS IN THE LABORATORY EVAPORATION TESTS.

The ratio of the penetration at 77° F. to that at 32° F. was always lower for the residue from exposure. The ductility at 77° F. of the residue from exposure was less than that of the asphaltic residue in all but two cases, and for these the ductility of both residues was 110+. In five cases the ductility at 34°-35° F. of the residue from exposure was greater than that of the asphaltic residue. In every case except two the percentage of material insoluble in naphtha was greater for the residue from exposure than for the asphaltic residue. In every case where there was an appreciable amount of material insoluble in carbon tetrachloride

and carbon disulphide, the percentages were much greater for the residues from exposure.

Figure 6 shows the development of free carbon, carbenes, and asphaltenes in laboratory and exposure tests for the samples that originally contained or finally developed carbenes in appreciable amounts. Samples 8, 23, and 27 developed only relatively small amounts of carbenes and the development of the insoluble constituents is shown only for sample 27. In figure 6 the volatile matter and the material insoluble in carbon disulphide, carbon tetrachloride, and 86° B. naphtha are plotted for the original materials and their residues.

TABLE 7.—Comparison of residues of approximately the same penetration from the asphaltic-residue test and from exposure

Sample identification	Asphaltic residue									Residue from exposure									
	Penetration		Pen. at 77° F. Ratio: Pen. at 32° F.	Softening point	Ductility, 5 cm per minute		B. Insoluble in 86° naphtha	In- organic matter soluble in CCl ₄	In- organic matter soluble in CS ₂	Penetration		Pen. at 77° F. Ratio: Pen. at 32° F.	Softening point	Ductility, 5 cm per minute		B. Insoluble in 86° naphtha	In- organic matter soluble in CCl ₄	In- organic matter soluble in CS ₂	Time of exposure
	At 77° F., 100 g., 5 sec.	At 32° F., 200 g., 60 sec.			At 77° F.	At 34°-35° F.				At 77° F., 100 g., 5 sec.	At 32° F., 200 g., 60 sec.			At 77° F.	At 34°-35° F.				
°F.	Centi- meters	Centi- meters	Percent	Percent	Percent	Percent	Percent	Percent	°F.	Centi- meters	Centi- meters	Percent	Percent	Percent	Weeks				
2	100	15	6.7	109	110+	0.0	35.2	2.16	0.16	86	34	2.5	108	63.0	1.5	30.2	9.23	2.74	5
3	105	15	7.0	107	110+	.0	32.5	1.50	.16	62	22	2.8	127	5.0	.0	31.3	11.29	4.46	10
4	107	18	5.9	108	110+	.0	25.5	1.62	.19	112	33	3.4	134	4.3	.0	29.3	7.41	2.46	10
5	105	28	3.8	113	110+	4.5	24.2	.12	-----	97	37	2.6	118	87.0	3.8	30.5	-----	-----	10
10	106	35	3.0	117	98	6.0	25.6	.12	-----	95	53	1.8	129	10.0	3.5	29.4	-----	-----	10
11	84	21	4.0	113	110+	.1	21.1	.03	-----	73	35	2.1	121	17.0	1.5	29.3	-----	-----	10
13	93	27	3.4	115	110+	1.0	22.7	.09	-----	89	39	2.3	119	40.0	3.5	27.3	-----	-----	10
14	93	13	7.2	110	110+	.0	31.1	.72	.16	80	21	3.8	112	97.0	.0	33.4	8.61	3.07	10
15	103	25	4.1	115	110+	4.5	23.1	.03	-----	105	47	2.2	117	53.0	3.5	30.3	-----	-----	10
18	88	27	3.3	120	110+	4.8	18.5	.15	-----	93	28	3.3	116	110+	4.3	18.8	.10	-----	15
20	102	21	4.9	114	110+	.8	19.5	.04	-----	84	28	3.0	120	72.0	3.0	27.2	.09	-----	15
22	113	23	4.9	109	110+	.0	14.5	.05	-----	112	28	4.0	111	110+	5.0	24.2	.10	-----	15

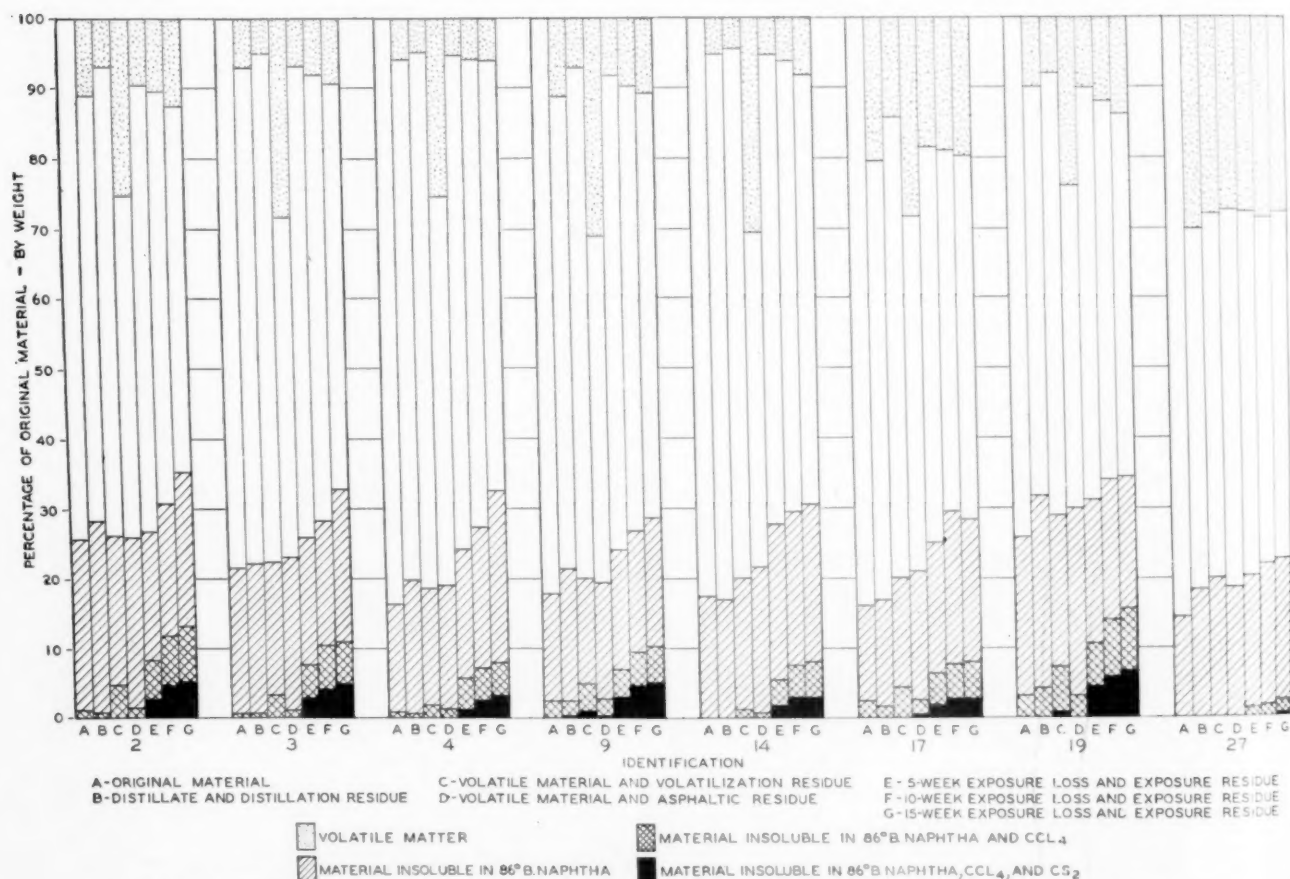


FIGURE 6.—COMPOSITION OF SELECTED MATERIALS AND THEIR RESIDUES AS DETERMINED BY SOLUBILITY TESTS.

All percentages are expressed in terms of the weight of the original material.

The solid portion in each vertical column represents free carbon or material insoluble in carbon disulphide. The remainder represents bitumen or material soluble in carbon disulphide. The double cross-hatched portion represents carbenes or bitumen insoluble in carbon tetrachloride, and the single and double cross-hatched portions together represent asphaltenes or bitumen

insoluble in 86° B. naphtha. The remainder of the column represents material soluble in 86° B. naphtha that, in the original material, includes the more volatile hydrocarbons vaporized and lost under test conditions, as shown by the dotted area. The materials soluble in 86° B. naphtha have been termed "malthenes" by Richardson.³ This designation, however, has not been generally accepted in the United States.

³ Clifford Richardson, *The Modern Asphalt Pavement*, p. 544 (2d ed.).

As shown by figure 6, the material insoluble in naphtha included carbenes and free carbon; in fact, in the residue from 15 weeks' exposure of sample 19 the material insoluble in naphtha contained 19 percent free carbon and 27 percent carbenes. The term asphaltenes also includes carbenes. Nellensteyn⁴ has stated that asphaltenes, carbenes, and free carbon all consist of the same matter, that is, dispersed carbon in decreasing states of protection. The so-called "protective bodies" he designates as "micelles." He states that when extracted asphaltenes are heated at a temperature of 527° F. they will be changed largely to free carbon, but a normal asphaltic material can be heated at 662° F. for a long period with little production of free carbon. The reason for this is that in normal asphalts the amount of protective bodies, or micelles, is such that the decomposition of part of them influences their protective qualities only slightly. Marcusson⁵ states that the material soluble in 86° B. naphtha (malthenes) is composed essentially of oily constituents and asphaltic resins. As the time of exposure increased, the percentage of asphaltenes, carbenes, and free carbon increased while the percentage of malthenes decreased.

In the 1932 investigation only those materials that had high specific gravities and initially contained some material insoluble in carbon disulphide with appreciable amounts insoluble in carbon tetrachloride developed carbenes either in laboratory or exposure tests. In the present study some of the materials such as samples 14 and 27, that originally had high solubilities in carbon disulphide and carbon tetrachloride, developed carbenes even during some of the laboratory tests. In those materials in which carbenes and free carbon were developed, it may be considered that the amount or the protective quality of the micelles was insufficient to prevent carbonization.

All of the rapid-curing products at the end of 15 weeks had developed residues containing about 0.5 percent of material insoluble in carbon disulphide and carbon tetrachloride. For all materials the solubility

in carbon disulphide was almost the same as the solubility in carbon tetrachloride, indicating the almost complete absence of carbenes.

MATERIAL INSOLUBLE IN NAPHTHA DEVELOPED MOST UNDER EXPOSURE AND LEAST IN THE DISTILLATION TEST

In figure 6, where all the percentages are expressed in terms of the weight of the original material, if the percentage of material insoluble in naphtha in any residue is divided by the percentage of material insoluble in naphtha in the original material and multiplied by 100, the result is the index of increase in material insoluble in naphtha. An index of 100 therefore indicates no change in the amount of material insoluble in naphtha. This index for all the samples is plotted in figure 7 and shows that generally there was an increase in material insoluble in naphtha during the various tests. Generally, this index was least for the distillation test and greatest for the 15-week exposure test. In the distillation test it ranged from 88 to 129.

While inaccuracies in testing may account for indexes of less than 100 in the distillation residues, it is possible for indexes actually to be below 100. Several samples of very viscous, slow-curing asphaltic material, really semisolid asphalts, were recently subjected to the distillation test. The materials did not yield any distillate but there was considerable loss on cooling. Nevertheless, the residues were softer than the original materials and contained less material insoluble in naphtha.

In the oven-loss test on 20-gram samples the index varied from 101 to 161 and in the asphaltic residue test it varied from 96 to 302. In the case of the asphaltic residue the index was low when the time of reduction was low, and for the cut-back materials the asphaltic residue test was the least severe of all tests, although there is not much difference between this and the distillation test. The index was high when the time of reduction was high, as indicated by sample 18 which took 420 minutes to come to 100 penetration and had an index of 302, and by sample 1 which took 280 minutes and had an index of 284. The index varied from 105 to 294 for 5 weeks' exposure, from 119 to 361 for 10 weeks' exposure, and from 133 to 376 for 15 weeks' exposure. Except for four samples, the index of increase was greater for 5 weeks' exposure than for any of the laboratory tests. In the exposure tests, products originally containing the highest percentages of material insoluble in naphtha generally had the smallest indexes of increase.

OLIENSIS TEST INVESTIGATED

An interesting development in the study of asphaltic materials is the test for determining heterogeneity. This test has been called the Oliensis or spot test. The method of making the test and the interpretation of the results were outlined in a paper read before the 1933 meeting of the American Society for Testing Materials. In making this test on the original material, one part by volume of the asphaltic material was treated with 5.1 parts by volume of a special naphtha at such a temperature that solution or dispersion was complete in 6 to 8 minutes. After cooling to room temperature and adding fresh naphtha to replace any losses, a drop of the mixture was allowed to drop on a filter paper (J. H. Munktells, No. 00). The appearance of the resulting stain varied from a uniformly colored spot, in-

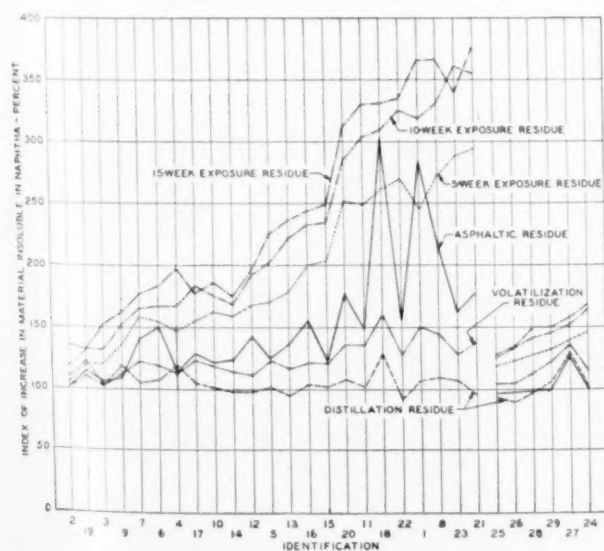


FIGURE 7.—INDEX OF INCREASE IN MATERIAL INSOLUBLE IN NAPHTHA IN THE VARIOUS RESIDUES.

⁴ Report by F. J. Nellensteyn and R. Loman, Sixth Congress, Permanent International Association of Road Congresses, first section, second question, paper 2-O.
⁵ See *Asphalts and Allied Substances* by Herbert Abraham, third edition, p. 755.

dicating complete dispersion, to stains in which the center was black and rough and surrounded by a lighter-colored ring. When an entirely uniform stain was obtained another test was made after the mixture had stood for 24 hours. The appearance of the central spot was taken as an indication of the degree of heterogeneity, and those materials that gave a uniform stain after standing 24 hours were considered as homogeneous.

The test is presumed to give an insight into the conditions of manufacture. The types of asphaltic products that should be expected to appear homogeneous are as follows:

1. Steam-refined residuals known to have been refined without serious cracking.
2. The bitumen of certain native asphalts.
3. Some types of slightly oxidized residuals from asphaltic-base crude oils.

The materials that should be expected to appear heterogeneous are:

1. Steam-refined residuals that have been overheated during the refining process.
2. Cracking-coil residuals.
3. Highly blown residuals

The test was initially developed and used to determine whether or not petroleum asphalts had been subjected to higher temperatures than usually occur in steam refining. It has been used by some States as an identification test for the control of liquid asphaltic materials. Recently the test procedure has been standardized by a group of Middle-Western States. In applying the test to slow-curing materials they state that if the material has less than 15 percent of distillate by volume at 680° F., the test may be made on the original material. For all other materials of the slow-, medium-, or rapid-curing classes the test shall be made on the distillation residue. The presence of the volatile distillate in these types of materials was thought to interfere with the sensitivity of the test. It will be noted that in the case of the slow-curing materials investigated the test may be run on 18 samples as received, since in only 5 cases was the percentage of distillate by volume at 680° F. more than 15 percent.

The identification of the character of the manufacturing process was the main object of this test, and for this purpose it should be made only on the finished products as they leave the refinery and not after they have been subjected to various laboratory heat tests. However, in order to determine if the character of the materials underwent a change during the laboratory tests and under exposure, the residue from the distillation test, the 20-gram oven-loss test, and the asphaltic-residue test, as well as the residues from the 5-week and 15-week exposures were subjected to the Oliensis test.

For the original materials 5.1 parts of naphtha by volume were mixed with 1 part of asphaltic material. For the various residues the volume of naphtha was kept constant and the weight of a unit volume of the original material minus the weight of volatile matter that occurred in producing the residue was used. Table 8 shows the ratio of naphtha to asphaltic material used for each sample. Only in the case of the asphaltic residue of sample 11 did the ratio of naphtha to asphaltic material exceed 7 to 1 by weight, the proportion that was being used by one State at the time these tests were made. It is thought that the variations in proportions of naphtha used in this work were not sufficiently wide to affect the character of the stains

TABLE 8.—Ratio of naphtha to asphaltic material in Oliensis test

Sample identification	Original material ¹ (by weight)	Distillation residue		20-gram loss residue (by weight)	Asphaltic residue (by weight)	Residue from exposure for—	
		By weight	By volume			5 weeks (by weight)	15 weeks (by weight)
1.....	4.10	4.21		4.12	5.69	4.10	4.10
2.....	3.74	4.20	5.83	4.01	5.00	4.13	4.27
3.....	3.74	4.01	5.51	3.92	5.22	4.00	4.13
4.....	3.79	4.01		3.96	5.07	3.99	4.02
5.....	3.96	4.58	6.04	4.33	5.61	4.37	4.47
6.....	4.06	4.38		4.26	6.49	4.35	4.42
7.....	4.11	5.37	6.90	4.30	6.37	5.15	5.24
8.....	4.13	4.51		4.27	6.32	4.20	4.22
9.....	3.83	4.29	5.86	4.11	5.55	4.16	4.28
10.....	4.10	5.28	6.84	4.92	6.22	4.90	4.96
11.....	3.91	5.22	6.94	4.61	7.35	4.91	5.03
12.....	4.10	4.99	6.38	4.57	6.68	4.75	4.84
13.....	3.91	4.31	5.68	4.17	5.86	4.21	4.33
14.....	3.73	3.92		3.89	5.36	3.93	4.06
15.....	4.03	4.77	6.18	4.42	6.54	4.44	4.65
16.....	4.07	4.37		4.22	6.56	4.26	4.37
17.....	3.83	4.82	6.80	4.52	5.18	4.70	4.76
18.....	4.13	5.02	6.42	4.95	5.52	5.07	5.10
19.....	3.65	4.05	5.76	3.97	4.73	4.06	4.23
20.....	4.12	4.78	6.05	4.53	5.95	4.65	4.76
21.....	4.06	4.77	6.07	4.55	6.52	4.64	4.74
22.....	4.05	4.48	5.70	4.38	5.76	4.42	4.52
23.....	3.90	4.41	5.86	4.29	6.44	4.25	4.40
24.....	4.06	5.40	7.40	5.22	5.27	5.22	5.24
25.....	4.17	5.48	7.24	5.38	5.34	5.36	5.36
26.....	4.16	5.44	7.14	5.36	5.31	5.37	5.34
27.....	3.99	5.72	7.98	5.54	5.50	5.52	5.54
28.....	4.14	5.24	6.90	5.20	5.15	5.17	5.18
29.....	4.11	5.38	7.29	5.24	5.24	5.24	5.24

¹ All original material 5.1 naphtha to 1 of sample by volume.

TABLE 9.—Character of original materials and residues as determined by the Oliensis test¹

Sample identification	Original material	Distillation residue	20-gram loss residue	Asphaltic residue	5-week exposure residue	15-week exposure residue
1.....	H	H	H	H	H	H
2.....	H	H	H	H	H	H
3.....	H	H	H	H	H	H
4.....	H	H	H	H	H	H
5.....	H	H	H	H	H	H
6.....	O	O	O	O	SH	H
7.....	SH	O	O	O	O	H
8.....	O	SH	SH	SH	H	H
9.....	H	H	H	H	H	H
10.....	O	SH	SH	O	H	H
11.....	H	H	H	H	H	H
12.....	O	O	SH	SH	H	H
13.....	H	H	H	H	H	H
14.....	H	H	H	H	H	H
15.....	O	SH	SH	SH	H	H
16.....	SH	SH	H	H	H	H
17.....	H	H	H	H	H	H
18.....	SH	SH	O	O	SH	H
19.....	H	H	H	H	H	H
20.....	SH	SH	O	O	H	H
21.....	O	O	O	O	H	H
22.....	O	O	O	O	H	H
23.....	H	H	H	H	H	H
24.....	H	H	H	H	H	H
25.....	O	O	O	O	SH	SH
26.....	O	SH	SH	O	SH	SH
27.....	H	H	H	H	H	H
28.....	O	O	O	O	SH	SH
29.....	H	H	H	H	H	H

¹ H = Heterogeneous; O = Homogeneous; SH = Slightly heterogeneous.

obtained. Table 9 gives a classification of the stains and figures 8, 9, and 10 show stains typical of those obtained with the various samples and their residues.

Since the results of the tests were based upon the appearance of the stain as interpreted by the observer, it is difficult if not impossible to distinguish between border-line materials or to express clearly the apparent degree of heterogeneity that may be indicated by the varying degrees of nonuniformity in the stain. The classification given in table 8 should be understood to mean that, in the judgment of the observers, the materials and their residues gave stains that appeared either entirely uniform throughout or were only slightly non-

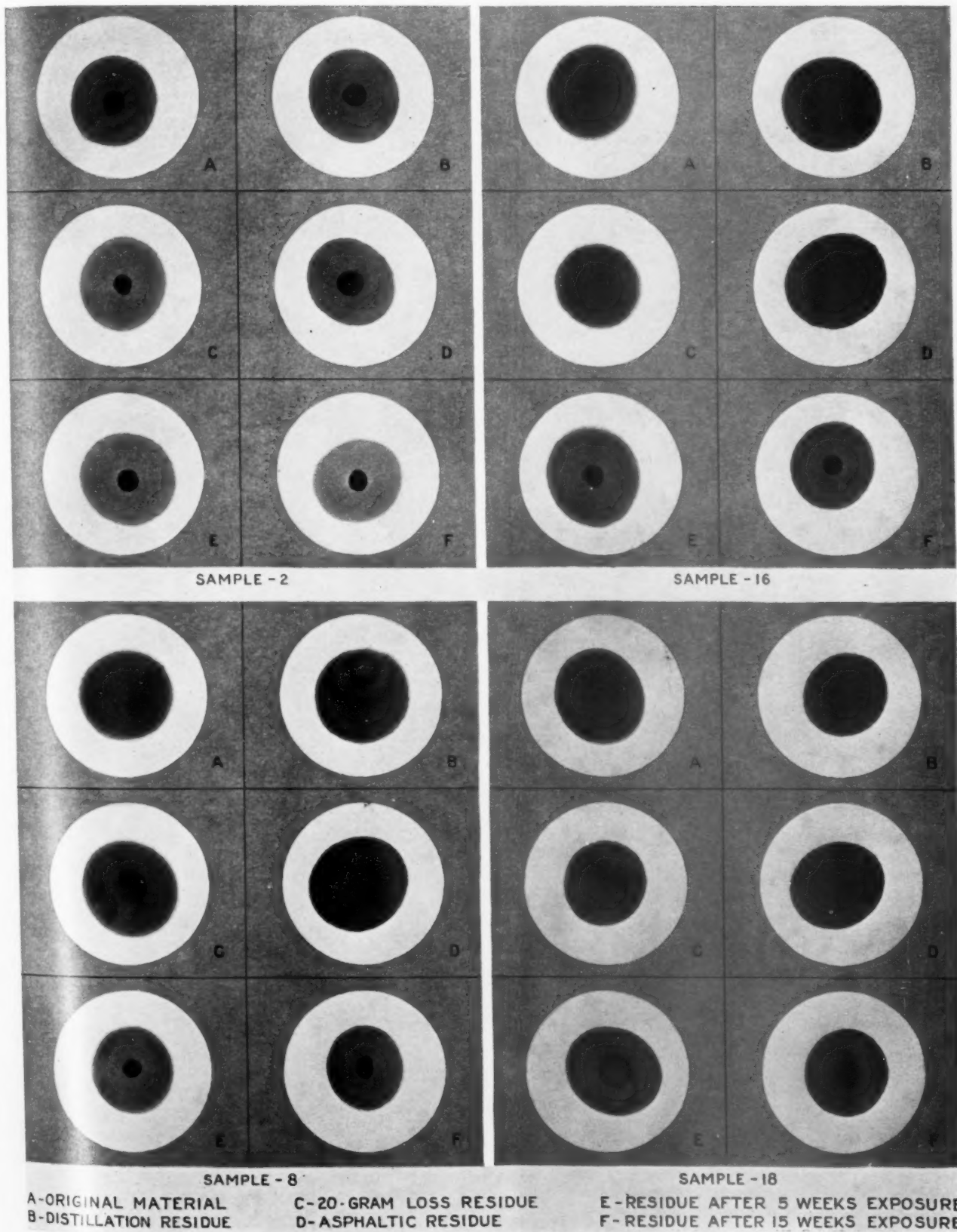


FIGURE 8.—TYPICAL OLIENSIS STAINS.

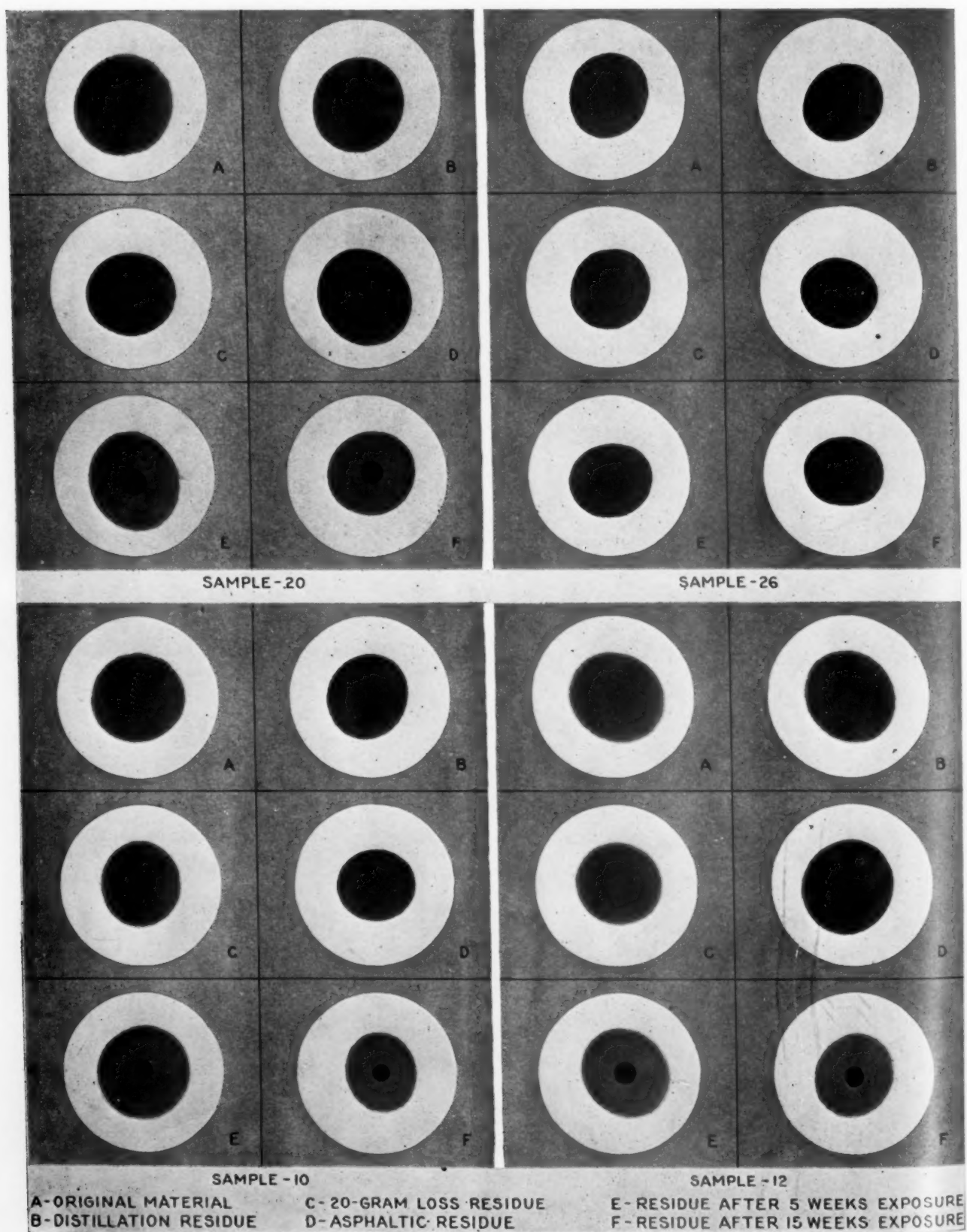


FIGURE 9.—TYPICAL OLIENSIS STAINS.

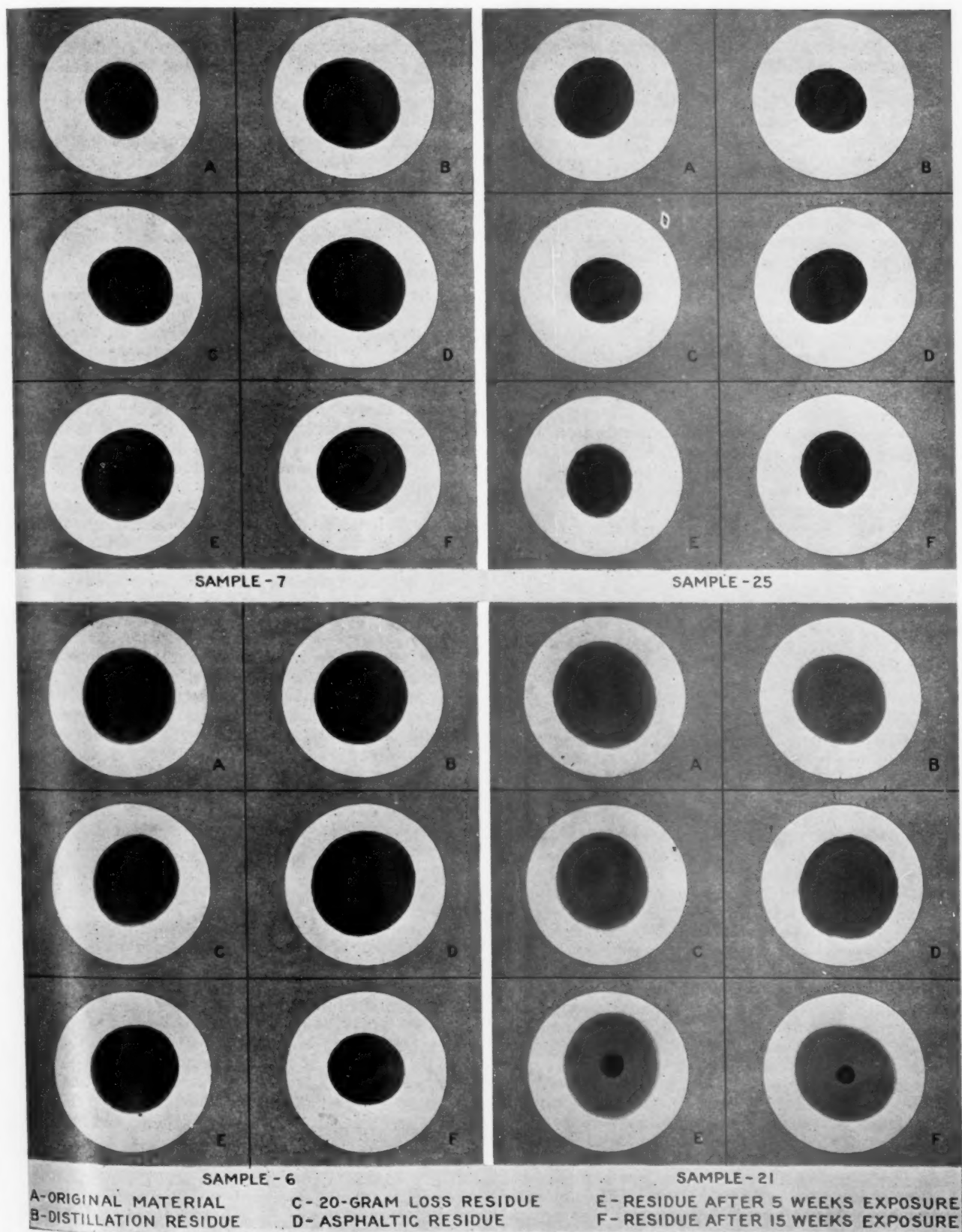


FIGURE 10.—TYPICAL OLIENSIS STAINS.

uniform, having a slightly darker, more-pronounced center, or else they had a definite dark to black center surrounded by a uniformly lighter-colored stain and were consequently classified as homogeneous, slightly heterogeneous, and heterogeneous, respectively. No attempt was made to indicate the extent or degree of heterogeneity other than to classify as slightly heterogeneous those materials and residues giving stains only slightly nonuniform.

ALL EXPOSURE RESIDUES FOUND TO BE HETEROGENEOUS

A study of the results of the Oliensis test shows that of the 29 materials, 10 were homogeneous, 4 slightly heterogeneous, and 15 definitely heterogeneous in their original state. All the residues from exposure were heterogeneous or slightly heterogeneous. Of the 10 materials appearing homogeneous in their original state, 5 developed homogeneous residues in all 3 laboratory loss tests, 2 developed slightly heterogeneous residues in all 3 tests, 2 developed slightly heterogeneous residues in the oven test and distillation test but not in the asphaltic-residue test, and 1 developed slightly heterogeneous residues in the oven test, and asphaltic-residue tests but not in the distillation test.

Of the 4 materials appearing slightly heterogeneous in their original state, 1 developed homogeneous residues in all 3 laboratory tests, 2 developed homogeneous residues in the asphaltic-residue and loss tests but remained heterogeneous in the distillation test, and one material remained heterogeneous in all tests. The 15 materials that appeared heterogeneous in their original state all developed heterogeneous residues.

A comparison of these materials according to the character of the stains produced in the laboratory and their behavior under laboratory and exposure conditions may be of interest. Nine of the materials, samples 2, 3, 4, 9, 14, 17, 19, 23, and 27, were heterogeneous originally and had heterogeneous residues. The photograph of sample 2 in figure 8 is typical. All of the above samples, except 23 and 27, showed microscopic flecks as shown in figure 2 when examined under the microscope. All except samples 14, 23, and 27 had appreciable amounts of material insoluble in carbon tetrachloride in the original sample. Samples 14, 23, and 27 developed carbenes at the end of 15 weeks of exposure and the other samples showed an increase in carbenes and free carbon.

All of the nine samples, except sample 23, had high percentages of material insoluble in naphtha. All had specific gravities greater than 1.01, except sample 27. This material has a very high specific gravity for a fluid cut-back and its behavior and characteristics place it in this group having high specific gravities. The asphaltic residues of the 9 samples showed the effect of changes in temperature, having low penetration at 32° F. and no ductility at 34°-35° F., although all had ductilities of over 110 at 77° F.

After 15 weeks' exposure all the materials were very hard, except sample 9, which, at the end of 5 weeks, had separated into two parts, one hard and brittle, the other soft and oily. It was impossible to flux these two parts so that, while having a float at 122° F. of over 1,000 seconds at the end of 5 weeks, it was impossible to get a penetration until the end of 15 weeks. When running the softening point test on the residues of this sample, after exposure, the material did not flow slowly to the bottom but dropped immediately at the temperature reported in table 5. Ductility tests were made on this

sample but the results are unimportant as no true ductility value was obtained. The residues after 15 weeks' exposure of samples 2, 3, 4, 17, 23, and 27 had very low ductility at 77° F. but samples 14 and 19 had ductilities at 77° F. of 97 and 110+, respectively. None of the residues had any ductility at 34°-35° F.

Three of the materials, 5, 11, and 13, gave Oliensis stains identical with those of the preceding samples. They did not, however, have as high specific gravities, were clear in the microscopic test, and did not, even at the end of 15 weeks, develop any carbenes. Their asphaltic residues had some ductility at 34°-35° F. and, at the end of 15 weeks, while having low ductility at 77° F., samples 5 and 13 had some ductility at 34°-35° F.

The remaining 17 materials all had low specific gravities, were clear in the microscopic test, did not have carbenes, and, with 1 exception, did not develop them. They showed various types of stains in the Oliensis test although all were heterogeneous at the end of 5 weeks. Their asphaltic residues varied in ductility and effect produced by changes in temperature as did their residues after exposure.

The stains of samples 1, 24, and 29, while resulting in photographs similar to those of the high-gravity materials, did not have the nuclei as raised or as rough as the stains of the materials with high specific gravities. The producers of the cut-back asphalt designated as sample 24 stated that their plant had no cracking equipment. This material, therefore, had evidently become heterogeneous in the refining process because of overheating, since the residue obtained in laboratory tests had good ductility, indicating that it had not been overblown. Sample 29 likewise was a cut-back asphalt with a ductile base. Sample 16 was slightly heterogeneous originally, produced slightly heterogeneous residues by distillation and volatilization, and produced a heterogeneous asphaltic residue. The asphaltic residues of these four samples were not especially affected by temperature change, having some ductility at 34°-35° F., although the asphaltic residues of samples 1 and 16 had comparatively low ductilities at 77° F. The residues of samples 1 and 16 after exposure had low ductility at 77° F. Those of samples 24 and 29 (cut-back asphalts) were very hard and consequently had very low ductility.

OLIENSIS TEST MORE SENSITIVE THAN MICROSCOPIC TEST IN DETECTING OVERHEATED MATERIALS

Samples 8 and 15 were homogeneous but all of their residues from laboratory tests were slightly heterogeneous. The asphaltic residue of sample 8 had relatively low ductility at 77° F. but that of sample 15 had good ductility. Both asphaltic residues had good ductilities at 34°-35° F. The residue of sample 15 after exposure had low ductility at 77° F. Sample 8, the only material of low specific gravity to develop carbenes, separated in the exposure test in the same manner as sample 9.

Samples 18 and 20 were both slightly heterogeneous but their asphaltic and loss-test residues were homogeneous. Their asphaltic residues and their residues after exposure had good ductility at 77° F. Although the asphaltic residue of sample 20 had a ductility of only three-fourths centimeter at 34°-35° F., the ductility of its residue after exposure, as well as the ductility of the two residues of sample 18, was good at 34°-35° F. Sample 18 developed a residue at the end of 5 weeks'

exposure that was the least heterogeneous of any of the residues from the slow-curing materials.

Samples 10 and 26 and their asphaltic residues were homogeneous. The asphaltic residues of both samples had good ductility at 77° F. and 34°-35° F. At the end of 15 weeks, sample 10 had low ductility at 77° F. but good ductility at 34°-35° F., while sample 26, a cut-back asphalt, was extremely hard and nonductile.

Sample 12 was homogeneous as was its residue after distillation. Its asphaltic residue had good ductility at 77° F. and 34°-35° F. and the residue after exposure had fair ductility at 77° F. and good ductility at 34°-35° F.

Sample 7 was slightly heterogeneous but all of its residues from laboratory tests were homogeneous. Its asphaltic residue was ductile at 77° F. but only slightly so at 34°-35° F. and its residue after 15 weeks of exposure had good ductility at 77° F. and at 34°-35° F.

Samples 6, 21, 22, 25, and 28 were homogeneous with homogeneous residues from laboratory tests. All of their asphaltic residues had good ductility at 77° F. and all except those of samples 21 and 22, the California residuals, had good ductility at 34°-35° F. At the end of 15 weeks, sample 6 was still fluid, while the cut-back asphalt samples 25 and 28 were very hard and nonductile. Samples 21 and 22 had good ductility at 77° F. and at 34°-35° F. although their asphaltic residues were nonductile at 34°-35° F.

It is readily apparent that the laboratory tests did not produce residues that gave stains in the Oliensis test radically different from the stains of the original materials. The behavior of the residues from exposure showed, as did the other tests, that outdoor exposure alters asphaltic materials far more than any of the laboratory heat tests. This was strikingly shown by the decidedly heterogeneous stains obtained with the residues from exposure, especially in the case of the materials originally homogeneous. It is not believed, however, that it is possible to predict the physical and chemical characteristics of the material after exposure from the results of the Oliensis test, whether made on the original material, the residues from laboratory tests or both. Residues having what are believed to be desirable qualities were obtained from both homogeneous and heterogeneous materials, although heterogeneous materials undoubtedly have a more pronounced tendency to carbonize and their slow-curing products generally develop a less-ductile residue.

For detection of materials that have been inadvertently or intentionally subjected to too high a temperature during the refining process, the Oliensis test seems to be more sensitive than the microscopic test. All of the materials that had the characteristics of overheated or cracked materials were heterogeneous in the Oliensis test but only seven of them showed microscopic flecks.

HUBBARD-FIELD STABILITY TEST USED TO MEASURE BONDING STRENGTH AND DEVELOPMENT OF BONDING STRENGTH UPON EXPOSURE

Cylinders were made according to the Hubbard-Field method and tested to determine the adhesiveness or bonding strength of the original material, the residue after distillation and the asphaltic residue, and the development of bonding strength by the original materials after exposure. The first series, for the determination of bonding strength, consisted of 3 sets of 3 cylinders each for each material. The first and second

sets contained 16.6 percent by volume of the original material and distillation residues respectively mixed with 83.4 percent of a standard sand. The third set contained the same percentage of asphaltic residue by weight as was contained in the cylinders made with the original materials that gave an almost constant percentage of bitumen by volume in the cylinders of this set. All cylinders of series 1 were tested immediately for stability at 77° F.

The second series of cylinders, for determination of the development of bonding strength, likewise consisted of 3 sets of 3 cylinders using the same aggregate used in the first series and the same percentage of the original materials by volume. These three sets were placed in the exposure boxes and subjected to the same exposure conditions as the thin films. One set was removed at the end of 5, 10, and 15 weeks. The cylinders were weighed before and after exposure and the loss in weight was expressed as a percentage of the bituminous material present in the cylinder as made. After weighing, the cylinders were tested for stability at 77° F.

For comparative purposes two additional sets of cylinders were made, using as a binder the amounts of distillation residue and asphaltic residue that would have been obtained if the bitumen in the cylinders containing the original material had been subjected to the distillation or asphaltic-residue test. The aggregate used was a Potomac River sand that had been separated on standard sieves and recombined to give the following grading:

	Percent
Passing no. 10, retained on no. 20.....	3.7
Passing no. 20, retained on no. 30.....	10.3
Passing no. 30, retained on no. 40.....	18.1
Passing no. 40, retained on no. 50.....	21.3
Passing no. 50, retained on no. 80.....	36.6
Passing no. 80, retained on no. 100.....	6.1
Passing no. 100, retained on no. 200.....	3.2
Passing no. 200.....	.7

This sand had a specific gravity of 2.666 and the voids in the mineral aggregate, determined on the compacted cylinders of both series, were 38 percent for the cylinders made with the original materials, 37.4 percent for the cylinders made with the distillation residue, and 36.9 percent for the cylinders made with the asphaltic residue.

The method of mixing and molding the cylinders was the same as that used in 1932. The results of the tests on the cylinders of series 1 and 2 are given in tables 10 and 11, respectively. All results are the averages of three tests.

The results of tests on the cylinders of series 1 are shown graphically in figure 11. The stability of the cylinders at 77° F. was plotted against the Furol viscosity at 122° F., and the results of the float test at 77° F. for the cylinders made with the original materials and against the float test results at 122° F. and the penetration at 77° F. for the cylinders made with the distillation residue. Since the asphaltic residues are all of approximately the same consistency, the stabilities were plotted for each sample independently.

Figure 11 shows that although the stability of the mixtures was roughly proportional to the consistency of the contained bitumen, materials having the same consistency as measured by viscosity at 122° F., float test at 77° F. and 122° F., and penetration at 77° F. had different stabilities. This was especially noticeable

TABLE 10.—Results of tests on series 1 cylinders

Sample identification	Original material			Distillation residue			Asphaltic residue	
	Stability at 77° F.	Float at 77° F.	Furoil viscosity at 122° F.	Stability at 77° F.	Float at 122° F.	Penetration at 77° F.	Stability at 77° F.	Penetration at 77° F.
	Pounds	Seconds	Seconds	Pounds	Seconds	Pounds	Pounds	
1	75	100	294	125	24	2,275	96	
2	125	52	303	400	60	3,975	100	
3	125	50	274	200	43	3,775	105	
4	150	63	404	250	45	3,600	107	
5	125	36	331	325	58	2,575	105	
6	75	21	335	150	21	2,275	99	
7	75	9	181	375	62	2,575	92	
8	100	60	277	100	27	2,100	100	
9	100	33	175	200	44	3,500	86	
10	75	11	197	475	87	2,425	106	
11	25	4	50	175	34	3,175	84	
12	100	13	182	200	38	2,450	97	
13	150	41	406	200	37	2,775	93	
14	125	49	278	150	36	3,675	93	
15	75	21	219	175	41	2,675	103	
16	75	38	217	100	25	2,225	109	
17	150	20	203	1,000	110	2,700	102	
18	125	9	199	400	60	1,900	100	
19	175	75	320	475	57	3,775	102	
20	100	46	300	350	50	2,325	104	
21	75	23	293	200	41	2,375	110	
22	150	60	768	325	56	2,325	113	
23	75	12	123	150	27	2,700	98	
24	75	14	241	3,925	76	2,950	101	
25	275	31	472	3,225	77	2,825	94	
26	150	35	351	3,025	84	2,725	95	
27	25	3	47	3,800	87	3,850	90	
28	200	19	303	3,775	65	3,725	97	
29	275	17	260	3,850	72	3,650	100	
30	125	39	350	3,475	81	3,000	104	
31	100	37	431	2,325	205	2,950	112	
32	75	25	331	1,500	298	3,075	101	

TABLE 11.—Results of tests on series 2 stability cylinders

Sample identification	Cylinders made with the original materials						Cylinders made with distillation residue		Cylinders made with asphaltic residue	
	Stability at 77° F.			Loss of bitumen			Stability at 77° F.	Theoretical loss of bitumen	Stability at 77° F.	Theoretical loss of bitumen
	When made	In 5 weeks	In 10 weeks	In 15 weeks	In 5 weeks	In 10 weeks				
	Lbs.	Lbs.	Lbs.	Lbs.	Per cent	Per cent	Lbs.	Per cent	Lbs.	Per cent
1	75	300	300	425	0	3	125	3	1,975	28
2	125	900	1,100	1,300	7	12	425	11	3,575	25
3	125	750	1,000	1,150	6	10	250	7	3,075	28
4	150	600	750	850	3	6	300	6	3,150	25
5	125	675	1,025	1,275	6	11	475	14	2,400	29
6	75	225	250	300	4	9	175	7	1,975	38
7	75	425	550	600	15	20	425	23	2,075	35
8	100	250	300	300	3	5	175	6	1,750	35
9	100	550	775	800	7	12	325	11	3,075	31
10	75	775	950	1,250	14	18	550	22	2,175	34
11	25	425	500	800	18	21	275	25	2,450	47
12	100	350	425	550	10	14	275	18	2,050	38
13	150	650	700	875	6	10	275	9	2,350	33
14	125	650	900	1,025	4	8	225	5	2,900	30
15	75	475	550	600	8	12	300	15	2,325	38
16	75	250	325	350	4	8	175	7	1,875	38
17	150	950	1,000	1,200	15	19	1,275	21	2,325	28
18	125	825	950	1,200	16	20	500	18	1,750	25
19	175	1,650	2,525	3,075	7	11	625	10	2,825	24
20	100	725	1,000	1,250	8	12	350	14	2,050	31
21	75	525	700	850	8	13	400	15	1,950	37
22	150	775	1,075	1,400	5	9	350	10	2,100	29
23	75	275	550	750	7	11	200	12	2,175	39
24	75	4,275	4,325	5,200	20	23	3,650	25	2,800	23
25	275	3,700	3,750	3,850	17	19	3,025	24	2,650	22
26	150	3,500	3,125	3,550	21	24	2,800	23	2,775	22
27	25	1,450	1,500	1,650	28	30	2,875	30	3,600	27
28	200	4,650	4,250	4,975	16	18	3,400	21	3,475	20
29	275	4,400	4,100	4,825	13	16	3,250	24	3,400	21
30	125	-----	-----	-----	-----	-----	3,175	26	2,700	24
31	100	-----	-----	-----	-----	-----	2,075	20	2,675	21
32	75	-----	-----	-----	-----	-----	1,425	22	2,775	25

in the results with the cylinders made with the asphaltic residue. Although all of these residues had approximately the same penetration, the stability of the cylinders varied from 1,900 pounds for sample 18 to 3,975 pounds for sample 2.

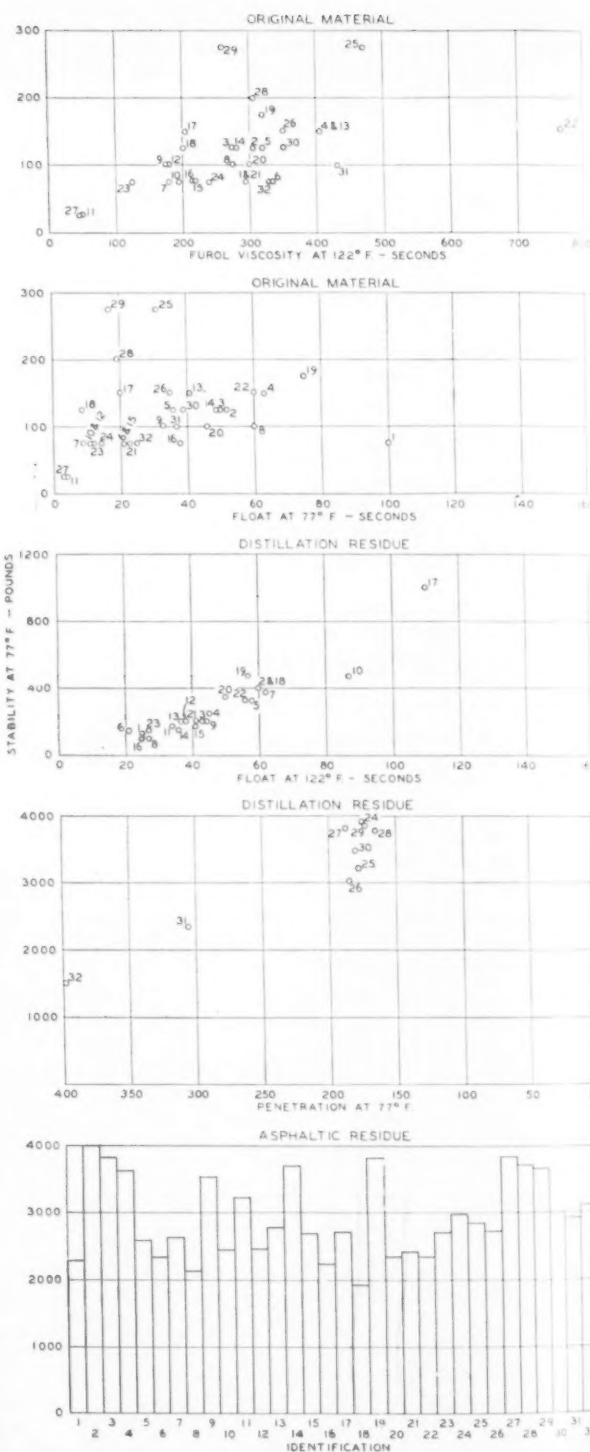


FIGURE 11.—Relation between the consistencies of original materials, distillation residues, and asphaltic residues and the stability at 77° F. of cylinders of series 1.

It is seen that the cylinders made with the asphaltic residues of samples 2, 3, 4, 9, 11, 14, and 19 all had stabilities of over 3,000 pounds. All of these materials were heterogeneous originally, all were materials of high specific gravity, and all except sample 11 contained or developed carbon and free carbon. Sample 11 had a relatively high specific gravity but did not develop

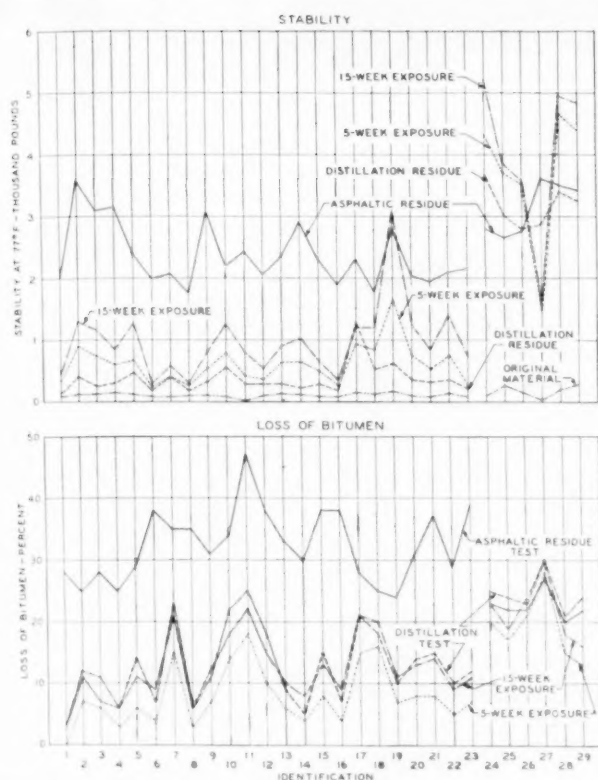


FIGURE 12.—COMPARISON OF LOSS OF BITUMEN AND STABILITY OF SERIES 2 HUBBARD-FIELD CYLINDERS.

carbenes. Samples 5, 7, 13, 15, 17, and 23 had stabilities between 2,500 and 3,000 pounds. Samples 17 and 23 were heterogeneous materials of high specific gravity that contained or developed carbenes. Samples 5 and 13 were heterogeneous materials of fairly high specific gravity but they did not develop carbenes, and samples 7 and 15 were materials of low specific gravity. All of the cut-back products had asphaltic residues giving stabilities of 2,500 pounds or over and five gave stabilities of 3,000 pounds or over.

Figure 12 shows the results of stability tests on the cylinders of series 2. The loss of bitumen in 5 and 15 weeks of exposure and the theoretical loss of bitumen in the cylinders made with the distillation and asphaltic residues were plotted for each sample. The stabilities at 77° F. for each sample were also plotted.

In this figure it is seen that, in the case of the slow-curing materials, although the loss of bitumen in the exposed cylinders was approximately the same as the loss in the distillation test, the exposed cylinders had greater stability than the cylinders made with the distillation residue except in the case of sample 17. This sample, even in 15 weeks, did not attain as high a stability as the cylinders made with the distillation residue. It is also seen that the loss in 15 weeks' exposure did not approach the loss in the asphaltic residue test and that the stability of the exposed cylinders did not approach the stability of the cylinders made with the asphaltic residue, except in the case of sample 19.

For the cut-back materials the indicated losses were probably in error due to unavoidable loss of volatile matter while mixing and molding the cylinders. The losses in 5 and 15 weeks of exposure probably should have been about the same as the losses in the distillation and asphaltic residue tests. The stabilities at 5 weeks

were higher than the stabilities of cylinders made with the asphaltic and distillation residues except in the case of sample 27.

SATISFACTORY CHECKS OBTAINED WITH RESULTS OF 1932 TESTS

After the exposure tests had been started, a question was raised concerning the use of plate glass covers for the exposure cabinets because it prevented the active ultra-violet rays from acting on the materials. Fused quartz glass not being available, Vita glass, which, after a short stabilization period, is guaranteed to permanently transmit an effective volume and combination of wave lengths of active ultra-violet light, was used to determine the effect of the passage of more active light. Duplicate sets of 10 of the slow-curing materials, 2 of the rapid-curing materials, and sample 27 and 3 new medium-curing materials, samples 30, 31, and 32, were exposed under both Vita and plate glass for 5- and 10-week periods. The materials were exposed in thin films and also admixed with the standard sand in the form of Hubbard-Field cylinders.

This exposure was started August 28, 1933. During the first 5-week period the average maximum air temperature was 80° F. and the number of sunlight hours was 266. During the 10-week period the average maximum air temperature was 73° F. and the number of sunlight hours was 512. The results of the tests on the thin films are given in table 12, and those on the Hubbard-Field cylinders in table 13.

As shown by tables 6 and 12, the materials did not lose as much nor get as hard in 10 weeks as they did in the original 5 weeks of exposure. This was due to the lower air temperature and also to the fact that the sun's rays striking at an oblique angle did not cause the material to get as hot as earlier in the summer. There was little if any difference between the materials exposed under the different types of glass. The samples exposed under Vita glass had a little more material insoluble in carbon disulphide and carbon tetrachloride and generally had a little more material insoluble in naphtha. Each solubility reported was the average of 3 or more tests. The results of the stability tests do not show that there was any difference between the two types of glass. The comparative study of the effectiveness of Vita glass and plate glass as cover for the exposure of the samples did not produce differences great enough to indicate their relative efficiency for this purpose.

As stated previously, four of the samples tested in 1932 were included in the 1933 work and the results obtained, as shown in table 14, were in remarkably close agreement with the previous tests. In the case of samples 17 and 19 the residues from the 1933 exposure tests had greater percentages of free carbon and carbenes than did residues from the 1932 exposure tests.

The results of the two sets of stability tests were not in such close agreement because the aggregate used in the 1933 tests was somewhat coarser than that used in 1932. In 1932, cylinders made with sample 17 were the only ones that, after 15 weeks' exposure, had a stability about the same as those made with the distillation residue. In 1933 the cylinders made with sample 17, after 15 weeks' exposure, had less stability than the cylinders made with the distillation residue. Cylinders of sample 19 that in 1932, after 15 weeks' exposure, had a stability approaching that of the asphaltic-residue cylinders, in 1933, after 15 weeks' exposure, had a stability higher than that of the asphaltic-residue specimens.

TABLE 12.—Results of tests on plate and Vita glass exposure residues

Sample identification	5 WEEKS' EXPOSURE										Vita glass									
	Plate glass										Vita glass									
	Loss	Float at 122° F.	Penetration		Softening point	Ductility at 5 cm per minute		Organic matter in-soluble in CS ₂	Organic matter in-soluble in CCl ₄	Insoluble in 86° B. naphtha	Loss	Float at 122° F.	Penetration		Softening point	Ductility at 5 cm per minute		Organic matter in-soluble in CS ₂	Organic matter in-soluble in CCl ₄	Insoluble in 86° B. naphtha
			At 77° F., 100 g., 5 sec.	At 32° F., 200 g., 60 sec.		At 77° F.	At 34°-35° F.						At 77° F., 100 g., 5 sec.	At 32° F., 200 g., 60 sec.		At 77° F.	At 34°-35° F.			
	Percent	Sec-onds			°F.	Centi-meters	Centi-meters	Percent	Percent	Percent	Percent	Sec-onds			°F.	Centi-meters	Centi-meters	Percent	Percent	Percent
1	-2.5	50								12.7	-3.3	46								13.4
2	6.7	113						0.82	6.86	20.9	4.5	92						1.22	6.85	29.2
13	4.5	61								18.4	2.2	54								17.8
14	2.2	53								24.8	1.6	47								23.4
17	12.4	140						1.08	4.09	26.8	12.4	156						1.28	4.64	28.3
18	16.0	110								12.2	14.9	104								12.9
19	6.4	105						1.68	9.63	32.6	5.6	98						2.00	10.22	33.6
20	7.9	103								17.2	6.5	96								17.0
21	8.0	53								14.1	5.7	49								13.7
23	6.0	52								19.2	5.6	50								19.2
26	21.5		37	18	143	8.5	0.0	.13	.33	27.3	21.5		33	16	147	10	0.0	.26	.15	27.6
27	27.4		46	14	124	17.0	.0	.19	.56	27.7	26.2		39	14	126	12	.0	.31	.64	28.0
29	18.5		40	17	141	11.0	.0			33.3	18.8		59	21	134	18	.0			31.7
30	22.6		38	16	136	18.0	.0			32.7	22.1		40	20	138	15	.0			33.5
31	17.6		61	18	124	110+	.0			29.7	17.3		52	17	126	110+	.0			30.8
32	17.7		122	24	106	110+	1			19.8	17.5		123	28	106	110+	5			20.8

Sample identification	10 WEEKS' EXPOSURE										Vita glass									
	Plate glass										Vita glass									
	Loss	Float at 122° F.	Penetration		Softening point	Ductility at 5 cm per minute		Organic matter in-soluble in CS ₂	Organic matter in-soluble in CCl ₄	Insoluble in 86° B. naphtha	Loss	Float at 122° F.	Penetration		Softening point	Ductility at 5 cm per minute		Organic matter in-soluble in CS ₂	Organic matter in-soluble in CCl ₄	Insoluble in 86° B. naphtha
			At 77° F., 100 g., 5 sec.	At 32° F., 200 g., 60 sec.		At 77° F.	At 34°-35° F.						At 77° F., 100 g., 5 sec.	At 32° F., 200 g., 60 sec.		At 77° F.	At 34°-35° F.			
	Percent	Sec-onds			°F.	Centi-meters	Centi-meters	Percent	Percent	Percent	Percent	Sec-onds			°F.	Centi-meters	Centi-meters	Percent	Percent	Percent
1	-2.5	54								0.14	14.1	-3.3	56							0.22
2	7.1	152						1.42		7.77	30.2	4.1	141					1.73		8.13
13	5.1	85								.04	19.6	3.2	71					.05		20.2
14	2.2	64						.50		3.16	23.9	2.3	65					.55		24.1
17	12.9	178						1.03		5.16	28.1	12.6	297					1.24		5.39
18	16.4	144								.12	12.9	16.1	154							.16
19	7.6	145						2.92		10.42	34.3	6.9	133					2.99		10.61
20	8.3	135								.07	19.4	8.0	125							.11
21	8.7	70								.09	15.0	8.7	74							.12
23	6.1	71								.10	20.4	6.1	84							.08
26	21.4		32	17	147	6.3	.0	.42		.36	28.0	21.5	30	16	149	6	.0	.31		.25
27	26.4		31	13	128	8.5	.0	.25		.73	29.9	26.7	26	13	130	7	.0	.30		.59
29	19.3		42	17	141	13.5	.0			.22	33.4	18.9	52	19	136	16	.0			.24
30	22.6		32	14	142	8.3	.0			.23	33.9	22.3	34	15	141	11	.0			.20
31	17.8		39	14	131	33.0	.0			.13	33.7	17.3	44	15	131	32	.0			.14
32	18.1		87	22	115	63.0	.0			.17	22.2	18.0	73	17	117	65	.0			.15

CONCLUSIONS

The results of this investigation substantiate most of the conclusions arrived at in the 1932 investigation, modify two of the conclusions, and indicate some new conclusions. The conclusions substantiated are:

1. Materials of high specific gravity and their residues are, in general, more susceptible to changes in temperature than materials of low specific gravity and their residues.

2. Hardening due to causes other than loss of volatile matter, and changes in inherent characteristics that may be attributed to oxidation, polymerization, and carbonization, occur to the greatest extent upon exposure and least during distillation.

3. The development of a ductile residue either in the asphaltic-residue test or, in the case of cut-back materials, in the distillation test, does not indicate that the material will develop a ductile residue upon exposure.

4. The bonding strength of the original materials and their residues is roughly proportional to their consistencies, but materials having the same consistency as measured by the present tests do not always give the same stability. The reasons for these differences in stability cannot be determined under the present methods of testing.

The following conclusions are somewhat modified from 1932:

TABLE 13.—Results of stability tests on cylinders exposed under plate and Vita glass

Sample identification	Loss of bitumen				Stability at 77° F.			
	In 5 weeks		In 10 weeks		In 5 weeks		In 10 weeks	
	Plate	Vita	Plate	Vita	Plate	Vita	Plate	Vita
	Percent	Percent	Percent	Percent	Pounds	Pounds	Pounds	Pounds
1	2	1	2	1	225	250	275	375
2	8	5	8	7	650	700	850	800
13	7	4	6	5	450	450	500	425
14	4	2	4	3	475	400	500	450
17	13	12	15	13	650	700	650	700
18	15	16	18	17	500	500	525	500
19	8	6	8	8	1,075	1,000	1,250	1,275
20	8	7	8	8	550	500	500	525
21	8	7	8	8	325	350	350	350
23	5	4	6	5	225	200	250	250
26	21	21	24	23	4,775	2,650	3,550	3,500
27	28	28	30	30	1,225	1,225	1,575	1,350
29	12	12	14	13	3,775	3,725	4,600	4,575
30	23	23	24	24	2,575	2,550	2,850	3,025
31	18	19	20	20	2,325	2,375	2,725	3,125
32	18	17	19	20	1,675	1,725	2,200	2,675

5. The relative rates of volatilization of the various materials can be anticipated most readily from the distillation curves. The different classes of material may be differentiated in the loss and asphaltic-residue tests, especially if the time of reduction to 100 penetration is considered. However, sharp distinctions in initial curing properties, that may be of importance in some types of construction, can be determined only from the distillation curves.

6. Carbonization generally occurs in materials that originally contain some material insoluble in carbon disulphide and carbon tetrachloride, but some materials with exceptionally high solubility in these solvents show a tendency to carbonize under both laboratory and exposure conditions.

The following conclusions are developed upon the basis of the data collected in 1933 only:

7. The Oliensis test is more sensitive than the microscopic test in the detection of materials that have been subjected to excessively high temperatures in manufacture. However, neither test seems definitely to distinguish products that will weather badly.

8. The use of Vita glass in place of plate glass for the cover of the exposure boxes did not materially change the results. However, because of the lateness in the year when these tests were started the results are considered inconclusive.

9. If like periods of the year are used for exposure, satisfactory check tests can be obtained with the exposure assembly used in these investigations.

Many of the laboratory heat tests have been criticized as producing conditions dissimilar to and more severe than service conditions. These investigations have shown that the physical and chemical characteristics generally believed to belong to unsatisfactory materials are developed upon exposure in many products that satisfactorily withstand laboratory testing. While it is possible, by the utilization of identification tests, to restrict materials to a limited number of sources or manufacturing processes, it is impossible to predict, with any degree of accuracy, the weather-resisting properties of the material thus obtained. It is believed that efforts should be directed to the modification of some of the present laboratory heat tests so that differences in the tendency of various materials to develop unsatisfactory residues may be recognized.

TABLE 14.—Comparison of 1932 and 1933 exposure tests

TESTS ON RESIDUES AFTER EXPOSURE

Test characteristic	Sample identification							
	17		18		19		20	
	1932	1933	1932	1933	1932	1933	1932	1933
Loss in 10 weeks, percent.....	18.3	18.7	18.5	17.7	13.3	12.0	13.1	11.5
Penetration at 77° F.....	41	36	132	152	40	44	95	136
Penetration at 32° F.....	17	14	37	43	9	10	30	41
Softening point, °F.....	163	170	106	105	121	121	115	112
Ductility at 77° F., centimeters.....	1.0	0.5	110+	110+	110+	110+	65.0	90.0
Ductility at 34°-35° F., centimeters.....	.0	.0	1.5	7.5	.0	.0	3.3	3.8
Insoluble in CS ₂ , percent.....	1.06	3.17			5.96	6.50		
Insoluble in CCl ₄ , percent.....	8.27	9.66	.10		14.94	16.02	.05	
Insoluble in 86° B. naphtha, percent.....	34.8	36.2	17.6	17.3	41.1	38.7	26.8	24.2
Loss in 15 weeks, percent.....	19.8	19.5	18.7	19.0	13.7	13.8	13.7	13.6
Penetration at 77° F.....	19	29	93	93	28	29	67	84
Penetration at 32° F.....	12	12	29	28	8	7	24	28
Softening point, °F.....	181	168	113	116	129	128	121	120
Ductility at 77° F., centimeters.....	0.3	0.5	86.0	115.0	110+	110+	42.0	72.0
Ductility at 34°-35° F., centimeters.....	.0	.0	4.0	4.5	.0	.0	1.3	3.0
Insoluble in CS ₂ , percent.....	2.50	3.30			6.17	7.50		
Insoluble in CCl ₄ , percent.....	9.11	9.88	.17	.10	15.61	18.15	.14	.09
Insoluble in 86° B. naphtha, percent.....	37.2	35.4	19.6	18.8	38.8	39.9	28.5	27.2

TESTS ON HUBBARD-FIELD CYLINDERS

Loss of bitumen:								
In 5 weeks, percent.....	15	15	18	16	8	7	9	8
In 10 weeks, percent.....	18	19	19	20	10	11	12	12
In 15 weeks, percent.....	21	21	21	20	14	11	13	13
Stability at 77° F.:								
Original cylinders, pounds.....	275	150	200	125	325	175	225	100
After 5 weeks' exposure, pounds.....	1,100	950	800	825	1,375	1,650	775	725
After 10 weeks' exposure, pounds.....	1,575	1,000	1,125	950	3,175	2,525	1,525	1,000
After 15 weeks' exposure, pounds.....	1,550	1,200	1,650	1,200	4,050	3,075	1,975	1,250
Series 2, distillation residue cylinders, pounds.....	1,475	1,275	750	500	800	625	500	350
Series 2, asphaltic residue cylinders, pounds.....	4,050	2,325	2,925	1,750	4,900	2,825	3,325	2,650

† Differences in stability probably are caused by differences in grading of the sand.

TABLE 12.—Results of tests on plate and Vita glass exposure residues

5 WEEKS' EXPOSURE

Sample identification	Plate glass										Vita glass									
	Loss	Float at 122° F.	Penetration		Softening point	Ductility at 5 cm per minute		Organic matter in-soluble in CS ₂	Organic matter in-soluble in CCl ₄	Insoluble in 86° B. naphtha	Loss	Float at 122° F.	Penetration		Softening point	Ductility at 5 cm per minute		Organic matter in-soluble in CS ₂	Organic matter in-soluble in CCl ₄	Insoluble in 86° B. naphtha
			At 77° F., 100 g., 5 sec.	At 32° F., 200 g., 60 sec.		At 77° F.	At 34° F., 35° F.						At 77° F., 100 g., 5 sec.	At 32° F., 200 g., 60 sec.		At 77° F.	At 34° F., 35° F.			
	Per-cent	Sec-onds			°F.	Centi-meters	Centi-meters	Per-cent	Per-cent	Per-cent	Per-cent	Sec-onds			°F.	Centi-meters	Centi-meters	Per-cent	Per-cent	Per-cent
1.	-2.5	50								12.7	-3.3	46								13.4
2.	6.7	113						0.82	6.86	29.9	4.5	92						1.22	6.85	29.2
13.	4.5	61								18.4	2.2	54								17.8
14.	2.2	53						.29	2.68	24.8	1.6	47						.47	2.73	23.4
17.	12.4	140						1.08	4.09	26.8	12.4	156						1.28	4.64	28.3
18.	16.0	110								12.2	14.9	104								12.9
19.	6.4	105						1.68	9.63	32.6	5.6	98						2.00	10.22	33.6
20.	7.9	103								17.2	6.5	96								17.0
21.	8.0	53								14.1	5.7	49								13.7
23.	6.0	52								19.2	5.6	50								19.2
26.	21.5		37	18	143	8.5	0.0	.13	.33	27.3	21.5		33	16	147	10	0.0	.26	.15	27.6
27.	27.4		46	14	124	17.0	.0	.19	.56	27.7	26.2		39	14	126	12	.0	.31	.64	28.0
29.	18.5		40	17	141	11.0	.0			33.3	18.8		59	21	134	18	.0			31.7
30.	22.6		38	16	136	18.0	.0			32.7	22.1		40	20	138	15	.0			33.5
31.	17.6		61	18	124	110+	.0			29.7	17.3		52	17	126	110+	.0			30.5
32.	17.7		122	24	106	110+	1			19.8	17.5		123	28	106	110+	5			20.8

10 WEEKS' EXPOSURE

1.	-2.5	54						0.14	14.1	-3.3	56							0.22	14.6
2.	7.1	152						1.42	7.77	30.2	4.1	141						1.73	8.13
13.	5.1	85						.04	19.6	3.2	71							.05	20.2
14.	2.2	64						.50	3.16	23.9	2.3	65						.55	3.45
17.	12.9	178						1.03	5.16	28.1	12.6	297						1.24	5.39
18.	16.4	144						.12	12.9	16.1	154							.16	13.9
19.	7.6	145						2.92	10.42	34.3	6.9	133						2.99	10.61
20.	8.3	135						.07	19.4	8.6	125							.11	19.0
21.	8.7	70						.09	15.0	8.7	74							.12	15.9
23.	6.1	71						.10	20.4	6.1	84							.08	21.3
26.	21.4		32	17	147	6.3	.0	.42	.36	28.0	21.5		30	16	149	6	.0	.31	.25
27.	26.4		31	13	128	8.5	.0	.25	.73	29.9	26.7		26	13	130	7	.0	.30	.59
29.	19.3		42	17	141	13.5	.0		.22	33.4	18.9		52	19	136	16	.0		.24
30.	22.6		32	14	142	8.3	.0		.23	33.9	22.3		34	15	141	11	.0		.20
31.	17.8		39	14	131	33.0	.0		.13	33.7	17.3		44	15	131	32	.0		.14
32.	18.1		87	22	115	63.0	.0		.17	22.2	18.0		73	17	117	65	.0		.15

CONCLUSIONS

The results of this investigation substantiate most of the conclusions arrived at in the 1932 investigation, modify two of the conclusions, and indicate some new conclusions. The conclusions substantiated are:

1. Materials of high specific gravity and their residues are, in general, more susceptible to changes in temperature than materials of low specific gravity and their residues.

2. Hardening due to causes other than loss of volatile matter, and changes in inherent characteristics that may be attributed to oxidation, polymerization, and carbonization, occur to the greatest extent upon exposure and least during distillation.

3. The development of a ductile residue either in the asphaltic-residue test or, in the case of cut-back materials, in the distillation test, does not indicate that the material will develop a ductile residue upon exposure.

4. The bonding strength of the original materials and their residues is roughly proportional to their consistencies, but materials having the same consistency as measured by the present tests do not always give the same stability. The reasons for these differences in stability cannot be determined under the present methods of testing.

The following conclusions are somewhat modified from 1932:

TABLE 13.—Results of stability tests on cylinders exposed under plate and Vita glass

Sample identification	Loss of bitumen				Stability at 77° F.			
	In 5 weeks		In 10 weeks		In 5 weeks		In 10 weeks	
	Plate	Vita	Plate	Vita	Plate	Vita	Plate	Vita
	Percent	Percent	Percent	Percent	Pounds	Pounds	Pounds	Pounds
1.	2	1	2	1	225	250	275	375
2.	8	5	8	7	850	700	850	800
13.	7	4	6	5	450	450	500	425
14.	4	2	4	3	475	400	500	450
17.	13	12	15	13	650	700	650	700
18.	18	16	18	17	500	500	525	500
19.	8	6	8	8	1,075	1,000	1,250	1,275
20.	8	7	8	8	550	500	500	525
21.	8	7	8	8	325	350	350	350
23.	5	4	6	5	275	200	250	250
26.	21	21	24	23	2,775	2,650	3,550	3,500
27.	28	28	30	30	1,225	1,225	1,575	1,350
29.	12	12	14	13	3,775	3,725	4,600	4,575
30.	23	23	24	24	2,575	2,550	2,850	3,025
31.	18	19	20	20	2,325	2,275	2,725	3,125
32.	18	17	19	20	1,675	1,725	2,200	2,675

5. The relative rates of volatilization of the various materials can be anticipated most readily from the distillation curves. The different classes of material may be differentiated in the loss and asphaltic-residue tests, especially if the time of reduction to 100 penetration is considered. However, sharp distinctions in initial curing properties, that may be of importance in some types of construction, can be determined only from the distillation curves.

6. Carbonization generally occurs in materials that originally contain some material insoluble in carbon disulphide and carbon tetrachloride, but some materials with exceptionally high solubility in these solvents show a tendency to carbonize under both laboratory and exposure conditions.

The following conclusions are developed upon the basis of the data collected in 1933 only:

7. The Oliensis test is more sensitive than the microscopic test in the detection of materials that have been subjected to excessively high temperatures in manufacture. However, neither test seems definitely to distinguish products that will weather badly.

8. The use of Vita glass in place of plate glass for the cover of the exposure boxes did not materially change the results. However, because of the lateness in the year when these tests were started the results are considered inconclusive.

9. If like periods of the year are used for exposure, satisfactory check tests can be obtained with the exposure assembly used in these investigations.

Many of the laboratory heat tests have been criticized as producing conditions dissimilar to and more severe than service conditions. These investigations have shown that the physical and chemical characteristics generally believed to belong to unsatisfactory materials are developed upon exposure in many products that satisfactorily withstand laboratory testing. While it is possible, by the utilization of identification tests, to restrict materials to a limited number of sources or manufacturing processes, it is impossible to predict, with any degree of accuracy, the weather-resisting properties of the material thus obtained. It is believed that efforts should be directed to the modification of some of the present laboratory heat tests so that differences in the tendency of various materials to develop unsatisfactory residues may be recognized.

TABLE 14.—Comparison of 1932 and 1933 exposure tests

TESTS ON RESIDUES AFTER EXPOSURE

Test characteristic	Sample identification							
	17		18		19		20	
	1932	1933	1932	1933	1932	1933	1932	1933
Loss in 10 weeks, percent.....	18.3	18.7	18.5	17.7	13.3	12.0	13.1	11.5
Penetration at 77° F.....	41	36	132	152	40	44	95	136
Penetration at 32° F.....	17	14	37	43	9	10	30	41
Softening point, °F.....	163	170	106	105	121	121	115	112
Ductility at 77° F., centimeters.	1.0	0.5	110+	110+	110+	110+	65.0	90.0
Ductility at 34°-35° F., centimeters.....	.0	.0	1.5	7.5	.0	.0	3.3	3.8
Insoluble in CS ₂ , percent.....	1.06	3.17	-----	-----	5.96	6.50	-----	-----
Insoluble in CCl ₄ , percent.....	8.27	9.66	.10	-----	14.94	16.02	.05	-----
Insoluble in 86° B. naphtha, percent.....	34.8	36.2	17.6	17.3	41.1	38.7	26.8	24.2
Loss in 15 weeks, percent.....	19.8	19.5	18.7	19.0	13.7	13.8	13.7	13.6
Penetration at 77° F.....	19	29	93	93	28	29	67	84
Penetration at 32° F.....	12	12	29	28	8	7	24	28
Softening point, °F.....	181	168	113	116	129	128	121	120
Ductility at 77° F., centimeters.....	0.3	0.5	86.0	115.0	110+	110+	42.0	72.0
Ductility at 34°-35° F., centimeters.....	.0	.0	4.0	4.5	.0	.0	1.3	3.0
Insoluble in CS ₂ , percent.....	2.50	3.30	-----	-----	6.17	7.50	-----	-----
Insoluble in CCl ₄ , percent.....	9.11	9.88	.17	.10	15.61	18.15	.14	.09
Insoluble in 86° B. naphtha, percent.....	37.2	35.4	19.6	18.8	38.8	39.9	28.5	27.2

TESTS ON HUBBARD-FIELD CYLINDERS

Loss of bitumen:								
In 5 weeks, percent.....	15	15	18	16	8	7	9	8
In 10 weeks, percent.....	18	19	19	20	10	11	12	12
In 15 weeks, percent.....	21	21	21	20	14	11	13	13
Stability at 77° F.:								
Original cylinders, pounds.....	275	150	200	125	325	175	225	100
After 5 weeks' exposure, pounds.....	1,100	950	800	825	1,375	1,650	775	725
After 10 weeks' exposure, pounds.....	1,575	1,000	1,125	950	3,175	2,525	1,525	1,000
After 15 weeks' exposure, pounds.....	1,550	1,200	1,650	1,200	4,050	3,075	1,975	1,250
Series 2, distillation residue cylinders, pounds.....	1,475	1,275	750	500	800	625	500	350
Series 2, asphaltic residue cylinders, pounds.....	4,050	2,325	2,925	1,750	4,900	2,825	3,325	2,050

¹ Differences in stability probably are caused by differences in grading of the sand.

MOTOR-FUEL CONSUMPTION, 1934

[Compiled for calendar year from reports of State authorities]

State	Gross amount reported by State ¹	Exempted from payment of tax ²		Shrinkage allowance, discounts, etc. ³	Gross amount assessed for taxation	Subject to refund of entire tax		Amount on which tax was earned	Classification of taxed motor fuel				Percentage change in fuel consumption from previous year ⁴
		Amount	Classes of use			By rate of tax			For high-way use ⁵	By use ⁶			
						At full rate	At reduced rates ⁴			Amount	Classes of use		
												Amount	
	1,000 gallons	1,000 gallons	Percent- age	1,000 gallons	1,000 gallons	Amount	Rate per gallon	1,000 gallons	1,000 gallons	Cents	Amount	Classes of use	
Alabama.....	154,977				154,977	10,280	NH, E	154,977					15.8
Arizona.....	*75,502	3,944		1	70,845	3,953	NH, E	60,565					13.0
Arkansas.....	*134,249	22,361	1	1	129,529	122,030	NH, C	125,576	17,102	6, 5, 4, 2	5,896	NH	4.3
California.....	1,356,386				1,320,685	24,013	NH	1,195,655					2.1
Colorado.....	*172,672	2,016	2	2	167,303	24,013	NH, D, E	143,290	68,909	4			7.6
Connecticut.....	*308,239	50,757	1	1	254,553	6,275	NH	248,658					3.4
Delaware.....	41,556				41,556	2,042	NH	30,514					5.2
Florida.....	246,387	7,564	(10)	(10)	235,698	235,698	NH	235,698					15.8
Georgia.....	243,823	2,203	(9)	(9)	239,435	239,435	NH, D, E	239,435	253	2 1/2	253	AV	13.7
Idaho.....	65,828	2,314			63,514	5,961	NH	57,300					25.5
Illinois.....	1,025,751				1,025,751	438,743	NH	970,874					4.6
Indiana.....	465,638				465,638	28,805	NH	438,743					7.8
Iowa.....	423,886	7,366	3	3	416,520	374,998	NH	374,998					20.0
Kansas.....	*378,781	87,330	2	2	291,451	283,876	NH	283,876					10.1
Kentucky.....	184,369				184,369	184,369	F	184,369					10.9
Louisiana.....	183,977		3	3	178,458	178,457	F	178,457	23	4			9.4
Maine.....	116,994	1,200			115,794	115,794	NH	115,794					8.7
Maryland.....	207,652	1,373			206,279	10,616	NH	195,663	4,870	1	4,870	NH	8.6
Massachusetts.....	*390,625	2,797			387,828	21,063	F, NH	366,735	1,702	3			3.8
Michigan.....	735,593				735,593	35,763	NH	698,831	1,149	1 1/2	1,149	AV	7.7
Minnesota.....	419,454	1,908	(10)	3	417,546	43,500	NH	364,042					8.3
Mississippi.....	130,156	2,180	2	2	127,976	125,429	NH	125,429	12,763	1	12,763	NH	16.5
Missouri.....	165,677	2,562	3	3	163,115	11,743	NH	149,402					5.4
Montana.....	*90,567		2	2	88,582	12,211	NH	73,271					33.2
Nebraska.....	*224,195	798	3	3	216,671	2,414	E, D	214,257					11.2
Nevada.....	*77,204	2,497			74,707	2,352	NH	22,355					28.5
New Hampshire.....	70,652				70,652	2,011	NH	68,641					16.8
New Jersey.....	*732,761	164,922			567,839	567,839	NH	567,839	112	2	112	B	3.9
New Mexico.....	*60,997	3,922	2	2	56,075	5,043	NH	51,134					12.9
New York.....	1,569,141	52,511	1	1	1,516,630	37,221	NH	1,464,242					1.3
North Carolina.....	284,214	2,540	1	1	279,797	21,485	NH	279,797	6,111	1	6,111	NH	11.2
North Dakota.....	96,875				96,875			75,390					17.6
Ohio.....	*1,303,642	239,900	3	3	1,063,742	673	D, R	1,031,157	120,943	1	120,943	NH	8.6
Oklahoma.....	*312,165	33,748	3	3	278,417	20,464	NH, E	270,432					7.5
Oregon.....	*165,978				165,978	20,464	NH, E	145,514	597	1	597	AV	6.7
Pennsylvania.....	1,136,343				1,113,629			1,113,629					8.7
Rhode Island.....	*233,167	124,303	2	2	108,864	6,030	NH, E	102,834					9.3
South Carolina.....	130,606				130,606	1,960	F	128,646					15.6
South Dakota.....	103,122	4,148	4	4	98,974			98,974					13.9
Tennessee.....	267,857		1	1	267,857	7	D, R	201,627	9,752	2	9,752	NH	8.7
Texas.....	893,802	9,824	1	1	883,978	84,133	NH	791,005					11.1
Utah.....	64,826		3	3	62,858			62,858					14.9
Vermont.....	48,550				48,550			48,550					10.0
Virginia.....	264,102				264,102	14,562	NH	249,540					12.6
Washington.....	290,778				290,778	21,591	NH	239,187					10.1
West Virginia.....	147,610				147,610	142,393	NH	142,393					15.8
Wisconsin.....	431,513		2 1/2	2 1/2	430,725	35,744	NH	384,981					1.5
Wyoming.....	44,111				44,111	44,111		44,111					25.5
District of Columbia.....	147,007	42,186	2	2	104,821	691	NH	103,129					-0.9
Total.....	17,220,567	882,618			16,136,137	681,656		15,454,481	244,346		162,446		7.5

The following symbols are used to designate certain classes or uses of motor fuel exempted from tax payment, subject to refund of the tax, or taxed at a lower rate:

F—Sales to Federal Government.
 P—(Public) sales to State, county, or municipal governments.
 E—Fuel exported to other States or countries.
 IC—Fuel moving through the State in interstate commerce.

NH—Uses other than for propelling motor vehicles on the highways.
 D—Fuel destroyed by fire, acts of God, etc.
 R—Routine refunds (overpayment, etc.).

B—Motor-boat use.
 D—Fuel destroyed by fire, acts of God, etc.
 R—Routine refunds (overpayment, etc.).

¹ In this column is given the total amount of motor fuel reported, prior to deduction of exempted fuel, allowance for shrinkage, etc., and amount subject to refund. Wherever possible, fuel exempted because of export or interstate movement, or to avoid duplication of tax payment, has been eliminated, in order that the total may represent as closely as possible the total consumption of motor fuel in the State. Starred items indicate that unknown amounts of export sales or fuel moving in interstate commerce are included in the total shown.

² A number of States failed to report exempted fuel. Symbols are given only where amounts are reported. Percentages are given only where amounts are reported. In some States the percentage is computed on the gross reported; in others on the net after deduction of exempted fuel. In some States the percentage is fixed; in others it is the maximum allowable.

³ In the case of Arkansas and Colorado, where the rate was changed during the year, the amounts taxed at the lower rates, 6 and 4 cents, respectively, are shown under this heading.

⁴ The purpose of this classification is to distinguish between the consumption of taxed motor fuel by motor vehicles operating on the highways, and consumption for other purposes. In the case of States which do not make this distinction, the classification is omitted.

⁵ These percentages are based on the amount taxed for highway use, except in the case of those States in which there is no classification by use. In those cases the percentage is based on the total amount on which tax was assessed. The total on which the Nation-wide percentage is based is 15,292,685,000 gallons. A decrease (District of Columbia only) is indicated by a minus sign.

⁷ Refunds on nonhighway use not allowed after Feb. 12, 1934.

⁸ 7,412,000 gallons at 6 cents prior to Feb. 13, 1934. Taxed at reduced rates at State borders: At 5 cents, 138,000 gallons; at 4 cents, 8,677,000 gallons; at 2 cents, 935,000 gallons.

⁹ Estimated by State.

¹⁰ Actual allowance reported; no fixed percentage.

¹¹ Rate of 4 cents per gallon applies to any gas-generating liquid having a flash point below 110° F. Additional 1-cent rate applies only to liquid fuels commonly used to propel motor vehicles or motors.

¹² Refunds are made on all nonhighway uses with the exception of fuel used in commercial motor boats.

¹³ Railroad usage.

¹⁴ Also fuel used in busses which pay a municipal or franchise tax.

¹⁵ A 3-cent tax is imposed on motor-vehicle fuel, and a 1-cent tax on all liquid fuels, including fuel oil and kerosene. The gross amount of liquid fuel reported was 1,303,642,000 gallons; the gross amount of motor-vehicle fuel reported was 1,254,786,000 gallons.

¹⁶ Tax is imposed on all liquid fuels, including fuel oil and kerosene, usable in internal-combustion engines.

STATE MOTOR-FUEL TAX EARNINGS, 1934

[Compiled for calendar year from reports of State authorities ¹]

State	Tax rate per gallon		Date of rate change	Gross tax assessed ⁴	Re-funds earned or paid ⁵	Net earnings on all motor-fuel taxed ⁶	Classification of tax earnings ¹				Other earnings in connection with motor-fuel tax ²						Grand total earnings
	On Jan. 1	On Dec. 31					By rate of tax			By use of fuel ³		Distributors' licenses	Dealers' licenses	Inspection fees	Other fees, etc.	Total	
							At full rate	At reduced rates ⁶		For highway use ⁷	For other uses						
								Rate per gallon	Amount								
	Cents	Cents		1,000 dollars	1,000 dollars	1,000 dollars	1,000 dollars	Cents	1,000 dollars	1,000 dollars	1,000 dollars	1,000 dollars	1,000 dollars	1,000 dollars	1,000 dollars	1,000 dollars	
Alabama.....	6	6		9,299		9,299	9,299									9,299	
Arizona.....	5	5		3,542	514	3,028	3,028			3,028				1	1	3,029	
Arkansas.....	6	6½	Feb. 3.....	8,118	254	7,864	7,047	6, 5, 4, 2	7,817	7,481	8,383					7,864	
California.....	3	3		39,621	3,661	35,960	35,960			35,960		10			10	35,970	
Colorado.....	4	4	Feb. 1 and Sept. 1.....	7,591	1,116	6,475	3,719	4	2,756	6,475						6,475	
Connecticut.....	2	2		5,099	126	4,973	4,973			4,973		(10)				4,973	
Delaware.....	3	3		1,246	61	1,185	1,185			1,185						1,185	
Florida.....	7	7		16,499		16,499	16,499					35			35	16,534	
Georgia.....	6	6		14,366		14,366	14,366									14,366	
Idaho.....	5	5		3,169	298	2,871	2,865	2½	6	2,865	6			1	1	2,872	
Illinois.....	3	3		30,772	1,646	29,126	29,126			29,126				5	5	29,131	
Indiana.....	4	4		18,626	1,076	17,550	17,550			17,550				20	20	17,570	
Iowa.....	3	3		12,114	864	11,250	11,250			11,250		1		4	5	11,255	
Kansas.....	3	3		8,516		8,516	8,516			8,516		5		76	37	8,634	
Kentucky.....	5	5		9,218		9,218	9,218							3	3	9,221	
Louisiana.....	5	5		8,923		8,923	8,922	4	1							8,923	
Maine.....	4	4		4,632	146	4,486	4,437	1	49	4,437	49					4,486	
Maryland.....	4	4		8,251	442	7,809	7,758	3	51	7,809						7,809	
Massachusetts.....	3	3		17,635	633	17,002	17,002			17,002						17,002	
Michigan.....	3	3		22,068	1,090	20,978	20,961	1½	17	20,961	17		4		2	20,984	
Minnesota.....	3	3		12,150	1,305	10,845	10,845			10,845			1	97		10,943	
Mississippi.....	6	6		7,519	631	6,888	6,760	1	128	6,760	128			137	137	7,025	
Missouri.....	2	2		9,810	235	9,575	9,575			9,575				98	8	9,681	
Montana.....	5	5		4,274	610	3,664	3,664			3,664						3,664	
Nebraska.....	4	4		8,667	97	8,570	8,570							85	1	8,656	
Nevada.....	4	4		988	94	894	894			894						894	
New Hampshire.....	4	4		2,826	80	2,746	2,746			2,746						2,746	
New Jersey.....	3	3		17,035	1	17,034	17,032	2	2	17,032	2		25	39	64	17,098	
New Mexico.....	5	5		2,809	252	2,557	2,557			2,557		9	9			2,575	
New York.....	3	3		45,041	1,117	43,927	43,927			43,927		58			58	43,985	
North Carolina.....	6	6		16,788	306	16,482	16,421	1	61	16,421	61			704	5	17,191	
North Dakota.....	3	3		2,906	644	2,262	2,262			2,262						2,262	
Ohio.....	4	4		38,982	1,364	37,618	36,409	1	1,209	36,409	1,209					37,618	
Oklahoma.....	4	4		10,817		10,817	10,817			10,817				4	4	10,821	
Oregon.....	5	5		8,299	1,047	7,252	7,246	1	6	7,246	6					7,252	
Pennsylvania.....	3	3		33,409		33,409	33,409							4	4	33,413	
Rhode Island.....	2	2		2,177	120	2,057	2,057			2,057			3			2,060	
South Carolina.....	6	6		7,836	117	7,719	7,719									7,719	
South Dakota.....	4	4		3,960	195	3,765	3,570	2	195	3,570	195					3,765	
Tennessee.....	7	7		14,114		14,114	14,114									14,114	
Texas.....	4	4		35,005	3,365	31,640	31,640			31,640				111	111	31,751	
Utah.....	4	4		2,514		2,514	2,514					1			1	2,515	
Vermont.....	4	4		1,942		1,942	1,942									1,942	
Virginia.....	5	5		13,205	728	12,477	12,477			12,477						12,477	
Washington.....	5	5		13,039	1,080	11,959	11,959			11,959						11,959	
West Virginia.....	4	4		5,905	209	5,696	5,696			5,696		2	5		7	5,704	
Wisconsin.....	4	4		16,829	1,430	15,399	15,399			15,399						15,399	
Wyoming.....	4	4		1,764		1,764	1,764									1,764	
Dist. of Columbia.....	2	2		2,077	14	2,063	2,063			2,063						2,063	
Detailed totals ¹⁸										434,634	2,056						
Grand totals.....		17 3.66		591,995	26,968	565,027	559,729		5,298			30	141	1,069	345	1,615	566,642

¹ See preceding table for gross gallons of motor fuel reported, exemptions, allowances, etc., gross gallons taxed, gallons subject to refund, net gallons taxed, and information regarding classes of use exempted, subject to refund, or taxed at lower rates.

² The purpose of this classification is to distinguish between the tax earnings on motor fuel sold for use in motor vehicles on the highways and tax earnings on motor fuel sold for other uses. In the case of those States that do not make this distinction, the classification is omitted.

³ Amounts less than \$500 not tabulated.

⁴ In the great majority of cases the assessments or earnings of the calendar year were reported. A few States reported the actual collections of the year, which lag the assessments by 1 to 2 months.

⁵ In most cases the refunds reported were those actually paid during the year, rather than refunds claimed on motor fuel purchased during the year. The error involved in deducting refunds paid from gross tax assessed tends to balance over an annual period. The refunds tabulated include both refunds of the entire tax and partial refunds.

⁶ In the case of Arkansas and Colorado, where the rate was changed during the year, the tax earnings at the lower rates, 6 and 4 cents, respectively, are shown under this heading.

⁷ Includes \$445,000 on 6-cent tax prior to Feb. 13, 1934, and amounts at reduced rates at State borders, as follows: At 5 cents, \$7,000; at 4 cents, \$347,000; at 2 cents, \$18,000.

⁸ Estimated by State.

⁹ Rate was 5 cents from Feb. 1 to Aug. 31, 1934.

¹⁰ Retail gasoline station licenses, \$45,000, included in report on motor-vehicle receipts.

¹¹ Includes distributors' licenses.

¹² Refunds are made on all nonhighway uses with the exception of fuel used in commercial motor boats. Earnings on motor-boat fuel (amount not reported) are included.

¹³ Includes \$138,560, earnings on special gasoline tax collected in Gulf Coast counties (Hancock, Harrison, and Jackson) for seawall protection, and \$1,559 in penalties, less \$2,629, refunds for notary fees.

¹⁴ Inspection fees on gasoline and kerosene; bulk of receipts on gasoline.

¹⁵ Includes dealers' licenses.

¹⁶ Classification by use includes 36 States and the District of Columbia.

¹⁷ Weighted average rate.

PUBLIC ROADS

CURRENT STATUS OF UNITED STATES PUBLIC WORKS ROAD CONSTRUCTION
AS PROVIDED BY SECTION 204 OF THE NATIONAL INDUSTRIAL RECOVERY ACT (1934 FUNDS) AND BY THE ACT OF JUNE 18, 1934 (1935 FUNDS)

CLASS 1.—PROJECTS ON THE FEDERAL-AID HIGHWAY SYSTEM OUTSIDE OF MUNICIPALITIES

AS OF JULY 31, 1935

STATE	APPORTIONMENTS		COMPLETED				UNDER CONSTRUCTION				APPROVED FOR CONSTRUCTION		BALANCE OF FUNDS AVAILABLE FOR NEW PROJECTS	
	Sec. 204 of the Act of June 18, 1934 (1935 Funds)	Act of June 18, 1934 (1935 Funds)	Total Cost	1934 Public Works Funds	1935 Public Works Funds	Mileage	Estimated Total Cost	1934 Public Works Funds	1935 Public Works Funds	Mileage	1934 Public Works Funds	1935 Public Works Funds	1934 Public Works Funds	1935 Public Works Funds
Alabama	3,034,753	2,139,921	5,174,674	3,460,795	1,713,879	356.2	2,147,104	1,460,140	1,264,071	109.8	20,073	195,768	6,795	430,180
Arizona	1,347,712	1,347,712	2,695,424	3,429,744	902,109	362.8	1,662,170	700,927	775,339	32.2	15,013	229,602	17,166	112,202
Arkansas	3,334,167	1,184,000	4,518,167	2,576,555	762,291	184.5	1,662,170	700,927	775,339	76.2	35,563	229,602	17,166	112,202
California	7,913,928	3,713,643	11,627,571	7,789,341	509,765	770.4	3,163,510	122,227	2,122,000	55.5	1,670	906,903	1,360	74,974
Colorado	3,437,265	2,484,504	5,921,769	3,151,026	275,088	262.3	1,131,322	295,088	516,992	30.5	11,350	3,063	12,768	5,774
Connecticut	1,504,213	607,500	2,111,713	1,109,195	1,002,018	46.6	1,131,322	295,088	516,992	17.1	3,063	3,063	3,063	5,774
Delaware	877,566	461,597	1,339,163	664,470	271,518	46.6	1,131,322	295,088	516,992	3.2	578	205,345	40	4,589
Florida	2,469,370	1,116,600	3,585,970	2,071,070	711,000	185.7	1,131,322	295,088	516,992	3.8	21,968	139,193	21,450	33,068
Georgia	5,045,592	2,556,745	7,602,337	3,371,140	761,641	336.9	1,131,322	295,088	516,992	86.1	6,394	185,422	139,193	74,165
Idaho	2,166,858	1,131,910	3,298,768	2,027,114	361,290	205.7	1,131,322	295,088	516,992	29.7	3,000	177,366	7,966	249,207
Illinois	4,408,827	2,408,778	6,817,605	2,391,815	145,249	40.3	1,131,322	295,088	516,992	1.1	7,951	235,515	11,350	73,045
Indiana	5,018,921	2,688,655	7,707,576	4,415,362	47,038	123.3	1,131,322	295,088	516,992	108.1	1,670	906,903	1,360	74,974
Iowa	5,027,870	2,212,361	7,240,231	5,508,137	394,520	331.3	2,002,083	49,000	1,555,141	108.1	13,700	11,350	11,350	294,000
Kansas	3,044,802	2,354,131	5,398,933	3,931,559	1,004,394	315.9	1,630,648	314,530	1,316,118	125.1	26,262	56,786	8,278	112,718
Kentucky	5,781,605	1,302,209	7,083,814	3,408,445	136,396	293.3	1,538,452	287,374	977,685	63.9	51,291	14,870	56,786	239,706
Louisiana	2,693,135	1,360,419	4,053,554	2,399,450	93,648	78.3	1,434,550	214,221	1,047,056	27.3	1,670	906,903	1,360	74,974
Maine	1,782,283	782,195	2,564,478	1,541,605	319,285	52.7	1,020,226	900,555	1,611,288	11.6	1,670	906,903	1,360	74,974
Maryland	1,782,283	782,195	2,564,478	1,541,605	319,285	52.7	1,020,226	900,555	1,611,288	11.6	1,670	906,903	1,360	74,974
Massachusetts	1,101,716	1,582,874	2,684,590	1,473,517	1,048,966	37.4	1,093,860	52,687	995,371	20.2	60,000	313,975	63	587,803
Michigan	6,091,533	3,226,284	9,317,817	5,670,820	5,213,895	960.4	3,440,799	782,950	2,657,849	145.9	59,087	141,276	11,350	25,571
Minnesota	4,561,011	2,533,733	7,094,744	6,200,552	1,740,016	960.4	3,440,799	782,950	2,657,849	145.9	59,087	141,276	11,350	25,571
Mississippi	3,469,337	2,832,182	6,301,519	2,682,221	688,779	292.9	2,503,432	723,191	1,780,241	144.1	23,588	601,637	60,337	265,571
Missouri	2,237,532	2,690,666	4,928,198	2,394,590	501,643	217.8	2,541,590	501,643	1,881,117	78.2	249,990	776,507	80,653	17,839
Montana	4,463,865	2,714,208	7,178,073	4,409,696	1,764,369	577.3	897,888	4,179	791,682	56.7	63,005	140,318	31,534	17,839
Nebraska	3,914,481	1,922,182	5,836,663	3,829,475	216,919	378.1	2,094,503	21,634	1,688,363	96.4	17,734	75,122	2,953	2,178
Nevada	2,909,387	1,750,356	4,659,743	2,666,269	703,341	332.7	804,013	204,618	260,113	147.2	17,734	75,122	2,953	2,178
New Hampshire	692,118	465,404	1,157,522	937,865	692,118	15.8	243,590	243,590	243,590	7.5	17,734	75,122	2,953	2,178
New Jersey	3,171,019	951,379	4,122,398	2,025,846	23,746	38.1	1,770,707	1,113,074	660,660	14.5	11,653	10,167	34,100	350,605
New Mexico	2,846,648	1,676,769	4,523,417	2,891,165	1,676,769	374.7	1,611,786	659,496	3,027,492	133.4	11,653	10,167	34,100	350,605
New York	10,465,672	3,673,231	14,138,903	9,094,794	5,044,109	235.2	7,084,590	1,153,792	3,027,492	133.4	9,000	18,900	237,906	169,949
North Carolina	1,930,365	1,469,404	3,400,769	3,684,500	423,710	644.2	1,435,381	663,896	665,944	137.4	91,438	322,565	321,703	518,146
North Dakota	2,277,254	1,462,744	3,740,000	2,641,138	315,993	1,165.4	1,536,460	74,016	356,496	177.6	183,824	151,500	41,167	677,656
Ohio	7,177,177	3,539,495	10,716,672	7,663,577	2,952,630	203.0	3,256,174	227,251	2,692,721	62.1	600	143,167	13,669	593,305
Oklahoma	4,608,399	2,342,590	6,950,989	5,005,046	616,560	330.6	2,000,652	348,296	1,436,273	70.7	6,016	143,167	4,272	146,493
Oregon	3,053,448	1,462,744	4,516,192	2,907,668	279,460	149.9	1,536,234	102,555	1,033,892	55.4	2,126	46,156	46,156	69,449
Pennsylvania	6,631,194	4,554,082	11,185,276	6,451,553	1,086,377	149.9	3,246,559	172,376	3,266,613	71.0	2,126	46,156	46,156	69,449
Rhode Island	979,367	474,772	1,454,139	899,627	168,738	25.6	391,854	79,740	295,634	9.3	18,520	51,578	35,811	10,200
South Carolina	2,729,563	944,794	3,674,357	2,539,651	214,683	253.4	1,447,575	198,818	1,041,045	49.6	11,766	153,378	35,811	10,200
South Dakota	3,096,719	1,523,621	4,620,340	2,532,418	474,277	685.0	1,395,130	809,160	192.4	11.0	27,475	96,985	46,881	124,290
Tennessee	4,246,309	2,105,494	6,351,803	3,999,811	577,343	201.7	1,508,309	176,447	1,236,361	53.3	2,195	183,445	67,756	109,304
Texas	11,586,643	6,808,253	18,394,896	13,416,341	1,503,032	1,161.1	4,713,720	240,731	4,472,989	338.2	25,675	694,311	47,4	434,718
Utah	2,857,205	1,066,345	3,923,550	3,170,271	572,595	274.7	1,480,185	31,000	344,179	28.4	17,000	17,000	7,787	135,271
Vermont	928,184	466,042	1,394,226	912,376	124,795	53.2	356,137	331,049	3,866	9.1	4,042	3,866	11,766	6,333
Virginia	3,731,207	1,916,176	5,647,383	3,850,819	632,668	153.1	1,447,575	198,818	1,041,045	69.7	92,935	153,378	27,475	87,887
Washington	3,057,934	1,553,206	4,611,140	2,859,879	474,277	114.9	1,286,948	227,922	938,447	15.1	92,935	153,378	27,475	87,887
West Virginia	2,013,405	1,140,167	3,153,572	2,301,844	359,092	86.5	496,108	42,643	453,465	14.8	27,869	47,122	2,093	350,389
Wisconsin	4,971,518	1,816,270	6,787,788	4,971,518	361,138	218.6	1,707,603	273,603	1,344,392	3.1	25,000	43,628	6,403	33,644
Wyoming	2,850,653	1,686,368	4,537,021	2,947,719	619,195	539.9	1,095,104	141,021	927,921	193.5	193.5	193.5	4,502	35,444
District of Columbia	1,693,344	594,778	2,288,122	859,029	566,724	19.3	1,430,996	1,126,894	99,424	20.3	20,973	183,932	18,793	315,362
TOTALS	185,235,296	93,535,660	278,770,956	165,756,136	24,036,147	13,962.3	80,941,786	16,595,744	55,080,948	3,693.4	1,139,370	6,565,135	1,733,986	8,233,430

CURRENT STATUS OF UNITED STATES PUBLIC WORKS ROAD CONSTRUCTION

AS PROVIDED BY SECTION 204 OF THE NATIONAL INDUSTRIAL RECOVERY ACT (1934 FUNDS) AND BY THE ACT OF JUNE 18, 1934 (1935 FUNDS)

CLASS 2.—PROJECTS ON EXTENSIONS OF THE FEDERAL-AID HIGHWAY SYSTEM INTO AND THROUGH MUNICIPALITIES

AS OF JULY 31, 1935

STATE	APPORTIONMENTS		COMPLETED				UNDER CONSTRUCTION				APPROVED FOR CONSTRUCTION				BALANCE OF FUNDS AVAILABLE FOR NEW PROJECTS			
	Sec 204 of the Act of June 18, 1934 (1934 Funds)	Act of June 18, 1934 (1935 Funds)	Total Cost	1934 Public Works Funds	1935 Public Works Funds	Mileage	Estimated Total Cost	1934 Public Works Funds	1935 Public Works Funds	Mileage	1934 Public Works Funds	1935 Public Works Funds	Mileage	1934 Public Works Funds	1935 Public Works Funds	Mileage	1934 Public Works Funds	1935 Public Works Funds
Alabama	2,389,928	1,004,961	3,394,889	1,872,058	119,375	49.1	165,135	165,135	165,135	23.4	29,878	122,470	3.1	24,965	500,084			
Arizona	756,982	306,191	1,063,173	738,883	86,475	35.1	232,930	232,930	232,930	23.4	101,076	311,035	5.7	4,857	61,883			
Arkansas	1,964,534	857,025	2,821,559	1,902,827	210,041	48.3	210,158	246,967	259,979	8.5				22,913	75,931			
California	4,213,986	2,219,360	6,433,346	5,358,835	847,032	65.8	2,680,138	356,840	1,070,374	8.0		194,200	5.5	7,422	107,784			
Colorado	1,718,635	190,000	1,908,635	1,937,435	169,411	40.4	11,289	11,289	11,289	1.6		194,613	1.3	35,346	20,589			
Connecticut	808,407	486,560	1,294,967	838,545	9,362	10.2	193,692	11,229	142,951	1.6					140,004			
Delaware	460,409	230,849	691,258	542,664	73,475	8.1	18,165	18,165	18,165	1.1		139,600	2.2	127	138,829			
Florida	1,459,648	501,500	1,961,148	1,839,561	102,427	21.4	166,128	2,324	165,804	1.4		31,607	1.2	17,554	94,953			
Georgia	2,764,620	1,276,373	4,040,993	2,414,917	146,915	74.8	701,177	275,754	426,423	18.2				215,987	693,715			
Idaho	1,097,429	324,185	1,421,614	1,222,156	27,081	20.9	190,673	1,056,345	189,445	5.5		35,000	3.2	46,349	69,560			
Illinois	7,341,315	2,339,160	9,680,475	6,622,453	84,400	26.5	2,121,458	1,056,345	1,065,092	11.1	47,260	630,357	3.1	12,455	516,050			
Indiana	4,287,090	2,846,958	7,134,048	3,615,513	108,596	70.4	2,881,628	1,056,345	1,825,283	29.0	142,115	189,445	5.5	27,544	282,342			
Iowa	2,614,472	1,311,000	3,925,472	2,303,093	101,635	54.9	1,408,378	508,465	898,913	11.3		137,700	1.2	39	282,465			
Kansas	2,522,401	1,432,949	3,955,350	3,007,648	317,503	46.8	1,059,769	528,141	531,628	9.9	48,919	10,940	1.2	30,000	213,325			
Kentucky	1,977,438	954,595	2,932,033	1,607,817	71,226	35.8	945,972	349,559	596,413	14.9	10,666	203,641	2.7	31,436	134,271			
Louisiana	1,708,877	744,560	2,453,437	948,432	73,980	27.5	1,107,913	896,730	209,181	3.7		138,253	2.7	242,658	462,515			
Maine	991,132	452,515	1,443,647	950,109	41,377	11.2	360,761	63,115	271,646	3.9				7,000	134,271			
Maryland	5,007,199	841,600	5,848,799	2,249,380	2,102,182	14.9	3,071,272	2,476,372	169,904	5.0		130,375	2.3	28,646	579,317			
Massachusetts	3,600,517	1,831,142	5,431,659	3,392,365	133,100	40.2	1,642,750	342,460	1,300,290	17.2	19,400	130,375	2.3	20,453	264,767			
Michigan	3,715,145	1,421,494	5,136,639	3,575,233	316,049	115.9	1,163,363	575,715	587,648	11.1	4,220	54,182	3.0	65,722	86,667			
Minnesota	1,794,669	394,022	2,188,691	1,166,445	114,966	37.0	817,590	626,097	191,493	23.4	36,943	18,447	2.5	48,163	100,163			
Mississippi	4,019,501	945,152	4,964,653	3,104,773	28,854	54.9	1,142,578	867,762	228,817	12.4		665,320	3.1	150,471	30,949			
Missouri	1,115,862	813,092	1,928,954	1,016,328	49,236	36.6	94,702	64,410	15,292	6.0	2,474	1,553	1.0	32,749	40,949			
Montana	1,977,840	991,091	2,968,931	2,498,314	544,276	46.3	296,070			4.5				41,657	146,743			
Nebraska	500,001	100,000	600,001	539,462	473,901	10.8	53,951			5.5				58,515	14,477			
Nevada	780,335	242,465	1,022,800	845,580	173,847	18.7												
New Hampshire	3,117,921	1,609,500	4,727,421	3,060,011	108,606	22.9	1,275,905	162,690	804,687	5.9		631,441	1.1	124,515	264,767			
New Jersey	1,674,758	529,566	2,204,324	1,776,171	180,097	40.2	69,432			1.7	1,010	181,263	1.1	77,794	96,014			
New Mexico	4,555,661	3,381,050	7,936,711	4,085,610	384,600	63.4	3,994,941	875,017	2,940,940	22.9		219,150	1.1	162,614	333,850			
New York	2,340,573	1,210,236	3,550,809	2,791,840	634,661	96.8	948,276	103,576	425,700	13.4		91,217	2.4	49,720	87,611			
North Carolina	1,451,112	734,741	2,185,853	1,341,415	92,515	54.5	166,059	99,644	48,185	9.9		130,561	16.4	22,386	30,949			
North Dakota	4,335,646	2,359,504	6,695,150	4,238,314	434,145	69.0	1,611,417	93,352	1,378,172	13.5		80,000	0.3	4,000	671,186			
Ohio	2,984,200	1,171,295	4,155,495	2,412,039	220,827	49.2	786,181	192,161	575,152	9.0		126,468	1.9	43,921	248,647			
Oklahoma	1,526,734	451,753	1,978,487	1,517,653	150,455	32.9	619,349	66,555	542,794	9.3		80,000	0.2	43,921	92,418			
Oregon	4,594,988	2,397,703	6,992,691	5,208,661	755,185	72.7	1,354,210	511,144	149,544	11.8		288,601	0.9	63,976	633,092			
Pennsylvania	579,625	285,760	865,385	556,591	36,001	8.0	106,759	156,274	109,759	9.9		125,412	1.1	60,634	144,004			
Rhode Island	1,304,791	448,000	1,752,791	1,136,674	23,620	38.2	420,472	244,409	291,547	12.8		125,412	1.1	136,716	144,004			
South Carolina	1,502,870	761,911	2,264,781	1,136,674	43,905	41.5	244,409	77,054	164,405	9.2		125,412	1.1	136,716	144,004			
South Dakota	2,123,156	1,121,789	3,244,945	2,209,866	264,518	28.3	586,283	291,632	144,601	7.6		115,946	0.7	246,369	246,675			
Tennessee	6,682,865	1,759,040	8,441,905	6,049,143	571,377	134.2	1,662,400	671,377	1,041,023	27.1		394,296	8.7	246,369	365,123			
Texas	778,426	533,173	1,311,599	812,144	97,066	20.8	550,233	125,130	425,103	13.5				246,369	2,386			
Utah	500,509	240,611	741,120	636,840	96,285	15.5	137,402	26,801	109,042	2.7		39,244	0.9	15,036	51,170			
Vermont	1,344,760	559,001	1,903,761	1,681,361	352,905	34.1	702,604	246,442	386,459	10.3		155,506	4.2	7,002	24,750			
Washington	1,577,460	776,605	2,354,065	1,362,048	35,313	42.5	412,559	53,569	359,190	6.8		125,412	1.1	1,846	144,004			
West Virginia	1,342,270	570,065	1,912,335	1,108,178	38,109	18.6	494,793	270,594	224,199	6.8		47,791	1.6	28,642	270,035			
Wisconsin	2,596,143	1,379,513	3,975,656	3,102,009	544,579	62.2	629,047	629,047	629,047	8.9		57,664	4.2	29,375	50,939			
Wyoming	1,125,332	977,416	2,102,748	977,416	2,784	22.5	142,381	147,667	14,132	3.4				6,474	12,500			
District of Columbia	946,446	141,051	1,087,497	877,332	111,051	6.5	250,164	250,164		0.2								
Hawaii																		
TOTALS	115,417,401	48,000,236	163,417,637	111,510,588	9,404,416	2,071.2	41,000,530	15,056,226	25,970,406	463.2	864,143	6,529,549	104.7	2,423,462	3,963,187			

PUBLIC ROADS

AS PROVIDED BY SECTION 204 OF THE NATIONAL INDUSTRIAL RECOVERY ACT (1934 FUNDS) AND BY THE ACT OF JUNE 18, 1934 (1935 FUNDS)

CURRENT STATUS OF UNITED STATES PUBLIC WORKS ROAD CONSTRUCTION

CLASS 3.—PROJECTS ON SECONDARY OR FEEDER ROADS

AS OF JULY 31, 1935

STATE	APPORTIONMENTS		COMPLETED				UNDER CONSTRUCTION				APPROVED FOR CONSTRUCTION			BALANCE OF FUNDS AVAILABLE FOR NEW PROJECTS		
	Sec 204 of the Act of June 18, 1934 (1934 Funds)	Act of June 18, 1934 (1935 Funds)	Total Cost	1934 Public Works Funds	1935 Public Works Funds	Mileage	Estimated Total Cost	1934 Public Works Funds	1935 Public Works Funds	Mileage	1934 Public Works Funds	1935 Public Works Funds	Mileage	1934 Public Works Funds	1935 Public Works Funds	
Alabama	\$ 2,012,452	\$ 1,064,960	\$ 2,016,508	\$ 1,803,119	\$ 196,710	143.0	\$ 822,610	\$ 187,765	\$ 634,865	52.6	\$ 120,940		10.6	\$ 41,549	\$ 104,426	
Arizona	259,421	298,012	557,433	550,262	357,034	86.5	644,037	501,989	563,989	45.2				23,461	77,009	
Arkansas	1,469,634	857,024	1,326,768	1,289,425	1,289,693	176.3	690,656	141,095	501,565	68.9	229,946		27.1	31,114		
California	3,460,340	1,999,203	4,050,207	3,154,463	216,000	180.1	1,833,841	323,497	1,510,344	56.4				1,700	155,951	
Colorado	1,512,532	871,502	2,601,784	1,606,632	593,169	141.5	2,221,880	110,000	300,140	39.2	262,527		9.4			
Connecticut	459,120	400,868	697,300	659,120	12,689	51.1	265,716	215,148	141,690	14.2					155,099	
Delaware	481,113	230,449	660,022	265,666	186,544	44.5	704,065	670,788	435,374	74.0				21,060	2,615	
Florida	1,202,816	1,041,943	1,650,616	1,279,796	331,460	84.5	1,000,362	704,065	435,374	74.0				109,711	765,690	
Georgia	2,350,973	1,278,373	1,660,592	1,399,474	55,108	185.3	2,847,261	2,856,472	2,847,261	27.0				27,032	179,044	
Idaho	1,131,565	623,450	1,670,519	1,094,530	398,344	106.8	5,646,397	274,972	3,382,895	266.5	473,156		7.8	13,958	334,402	
Illinois	5,702,071	4,232,273	3,633,423	3,495,503	91,422	205.5	3,781,598	2,856,472	1,021,626	50.2				35,227	42,347	
Indiana	2,413,358	1,590,000	2,789,645	2,331,399	431,850	145.7	1,555,821	119,252	1,144,910	246.0						
Iowa	2,522,461	1,310,595	2,326,666	2,131,401	199,358	294.7	1,468,690	309,453	1,181,217	171.8						
Kansas	1,527,956	1,557,503	2,175,313	1,772,049	275,359	241.3	1,341,513	46,365	1,211,964	167.8						
Kentucky	1,426,479	812,893	1,198,264	1,056,013	101,476	50.5	793,630	209,970	589,699	38.7						
Louisiana	842,473	446,052	1,388,314	842,404	390,202	101.1	1,512,506	37,523	51,608	3.4				374	18,305	
Maine	891,132	1,067,934	573,933	895,479	60,271	68.5	37,523	16,123	359,120	15.3				30	3,102	
Maryland	425,185	920,000	477,470	469,741	101,440	35.2	523,663	209,207	1,467,417	15.9						
Massachusetts	2,376,445	1,613,142	3,275,530	2,937,940	101,440	312.0	1,872,644	1,431,444	701,299	89.1				18,441	82,567	
Michigan	1,744,569	3,94,023	1,266,237	1,248,797	16,500	143.1	546,417	460,544	85,462	40.9				40,292	15,866	
Minnesota	2,923,273	2,363,322	3,198,127	2,686,002	431,784	673.2	1,928,533	233,531	1,707,003	359.5				51,113	95,132	
Mississippi	1,593,337	942,434	2,384,231	1,740,533	275,595	283.4	311,474	405,493	1,707,003	359.5				37,621		
Montana	931,091	2,472,284	1,472,284	1,472,284	402,603	329.2	405,493	12,000	180,528	28.0				8,443		
Nebraska	692,000	1,586,093	1,111,353	1,111,353	402,603	329.2	405,493	12,000	180,528	28.0				8,443		
Nevada	1,717,326	570,253	770,253	441,386	74,757	28.3	206,460	29,000	174,595	6.4				2,683		
New Hampshire	55,099	460,000	56,538	55,099	371,366	5.5	107,626	36,911	107,626	1.7				170,412		
New Jersey	1,272,159	735,425	1,656,584	1,327,536	298.0	119.5	386,791	36,911	3,034,790	41.4				65,732		
New Mexico	3,506,768	3,693,000	3,027,536	3,027,536	572,549	119.5	4,153,510	515,500	3,034,790	257.0						
New York	2,380,573	1,700,340	2,865,766	2,176,794	687,943	319.4	1,023,082	160,265	868,777	101.6				134,315		
North Carolina	1,451,112	734,742	1,817,766	1,410,948	46,074	387.3	201,989	134,295	50,713	54.6				100,391		
North Dakota	3,871,146	1,866,253	4,172,803	3,173,018	109,850	322.6	1,416,066	49,860	1,318,253	89.6				15,304		
Ohio	2,309,199	1,171,295	2,402,553	2,052,553	70,199	273.4	1,239,063	291,646	694,824	56.1				15,384		
Oklahoma	1,526,722	777,096	2,115,745	1,494,861	447,577	167.9	402,753	15,556	322,414	19.0				28,216		
Oregon	7,344,822	2,639,003	6,718,399	6,207,932	341,038	555.1	3,353,144	1,093,609	2,160,362	198.5				70,269		
Pennsylvania	439,716	299,040	449,746	439,716	127,475	33.1	212,563	263,518	212,563	6.7				15,117		
Rhode Island	1,364,791	1,342,000	1,234,563	1,057,572	127,475	130.1	1,490,297	174,741	1,148,209	107.5				43,080		
South Carolina	1,502,670	761,911	1,498,798	1,355,467	131,844	436.9	573,960	174,741	399,219	139.5				14,619		
Tennessee	2,123,155	1,075,748	2,004,259	1,771,569	165,906	146.9	837,013	263,714	567,299	37.6				2,661		
Texas	6,012,318	3,610,460	5,952,792	5,952,792	694,201	872.7	2,911,721	34,517	2,865,130	170.8				63,978		
Utah	1,048,677	533,172	1,697,318	994,955	271,173	217.4	407,861	34,517	295,000	46.8				25,209		
Vermont	438,860	241,354	694,760	83,862	165,082	47.9	80,229	74,410	74,410	5.3				3,019		
Virginia	1,236,170	893,188	1,716,601	1,251,099	105,265	216.3	653,915	511,717	507,790	22.7				18,141		
Washington	1,683,673	776,603	1,080,673	1,080,673	267,589	102.1	515,950							57,607		
West Virginia	1,114,559	570,083	799,957	799,957	308,675	43.5	664,954	318,088	346,766	28.0				47,546		
Wisconsin	2,431,220	1,741,354	2,644,467	2,431,220	238,408	56.7	1,661,341	1,336,317	1,336,317	7.3				5,338		
Wyoming	1,125,332	571,928	1,200,239	1,047,457	135,376	156.8	490,373	75,284	314,598	115.2				2,590		
District of Columbia	972,084	792,791	1,160,963	971,729	189,234	10.4	393,065			2.5				295		
Hawaii	377,718	351,000	178,209			4.3								161,653		
TOTALS	93,147,363	58,913,402	81,877,442	11,928,337	10,024.3		46,503,438	9,859,427	36,564,187	3,734.3	513,088	4,440,494	393.0	897,466	5,019,424	

PUBLICATIONS of the BUREAU OF PUBLIC ROADS

Any of the following publications may be purchased from the Superintendent of Documents, Government Printing Office, Washington, D. C. As his office is not connected with the Department and as the Department does not sell publications, please send no remittance to the United States Department of Agriculture.

ANNUAL REPORTS

- Report of the Chief of the Bureau of Public Roads, 1924.
5 cents.
- Report of the Chief of the Bureau of Public Roads, 1927.
5 cents.
- Report of the Chief of the Bureau of Public Roads, 1928.
5 cents.
- Report of the Chief of the Bureau of Public Roads, 1929.
10 cents.
- Report of the Chief of the Bureau of Public Roads, 1931.
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- Report of the Chief of the Bureau of Public Roads, 1932.
10 cents.
- Report of the Chief of the Bureau of Public Roads, 1933.
- Report of the Chief of the Bureau of Public Roads, 1934.

DEPARTMENT BULLETINS

- No. 136D . . Highway Bonds. 20 cents.
- No. 347D . . Methods for the Determination of the Physical Properties of Road-Building Rock. 10 cents.
- No. 583D . . Reports on Experimental Convict Road Camp, Fulton County, Ga. 25 cents.
- No. 1279D . . Rural Highway Mileage, Income, and Expenditures, 1921 and 1922.

TECHNICAL BULLETINS

- No. 55T . . . Highway Bridge Surveys. 20 cents.
- No. 265T . . . Electrical Equipment on Movable Bridges.
35 cents.

MISCELLANEOUS CIRCULARS

- No. 62MC . . Standards Governing Plans, Specifications, Contract Forms, and Estimates for Federal-Aid Highway Projects. 5 cents.

MISCELLANEOUS PUBLICATIONS

- No. 76MP . . The results of Physical Tests of Road-Building Rock. 25 cents.
- Federal Legislation and Regulations Relating to Highway Construction. 10 cents.
- Supplement No. 1 to Federal Legislation and Regulations Relating to Highway Construction.
- No. 191 . . . Roadside Improvement. 10 cents.
- The Taxation of Motor Vehicles in 1932. 35 cents.

REPRINT FROM PUBLIC ROADS

- Reports on Subgrade Soil Studies. 40 cents.

Single copies of the following publications may be obtained from the Bureau of Public Roads upon request. They cannot be purchased from the Superintendent of Documents.

SEPARATE REPRINT FROM THE YEARBOOK

- No. 1036Y . . Road Work on Farm Outlets Needs Skill and Right Equipment.

TRANSPORTATION SURVEY REPORTS

- Report of a Survey of Transportation on the State Highway System of Ohio (1927).
- Report of a Survey of Transportation on the State Highways of Vermont (1927).
- Report of a Survey of Transportation on the State Highways of New Hampshire (1927).
- Report of a Plan of Highway Improvement in the Regional Area of Cleveland, Ohio (1928).
- Report of a Survey of Transportation on the State Highways of Pennsylvania (1928).
- Report of a Survey of Traffic on the Federal-Aid Highway Systems of Eleven Western States (1930).

A complete list of the publications of the Bureau of Public Roads, classified according to subject and including the more important articles in *PUBLIC ROADS*, may be obtained upon request addressed to the U. S. Bureau of Public Roads, Willard Building, Washington, D. C.

CURRENT STATUS OF UNITED STATES PUBLIC WORKS ROAD CONSTRUCTION

AS PROVIDED BY SECTION 204 OF THE NATIONAL INDUSTRIAL RECOVERY ACT (1934 FUNDS) AND BY THE ACT OF JUNE 18, 1934 (1935 FUNDS)

SUMMARY OF CLASSES 1, 2, AND 3.

AS OF JULY 31, 1935

STATE	APPORTIONMENTS		COMPLETED				UNDER CONSTRUCTION				APPROVED FOR CONSTRUCTION			BALANCE OF FUNDS AVAILABLE FOR NEW PROJECTS		
	Sec. 204 of the Act of June 18, 1934 (1934 Funds)	Act of June 18, 1934 (1935 Funds)	Total Cent	1934 Public Works Funds	1935 Public Works Funds	Mileage	Estimated Total Cost	1934 Public Works Funds	1935 Public Works Funds	Mileage	1934 Public Works Funds	1935 Public Works Funds	Mileage	1934 Public Works Funds	1935 Public Works Funds	
Alabama	8,370,133	4,259,842	10,644,744	7,135,888	495,966	250.3	3,756,109	1,110,945	2,222,031	185.9	49,951	447,178	29.4	73,309	1,034,647	
Arizona	5,211,960	2,641,925	6,401,137	4,381,509	1,066,882	466.5	1,874,076	1,174,322	1,496,161	78.5	15,039	770,983	39.1	39,090	138,892	
Arkansas	6,744,335	3,428,049	7,413,454	5,411,547	965,678	409.1	2,823,623	1,096,989	1,503,695	153.6	140,639	1,353,531	37.7	93,159	188,133	
California	15,507,394	7,932,206	20,669,040	14,793,908	1,672,794	616.3	7,677,909	822,064	4,557,069	180.0		1,353,531	37.7	10,482	338,709	
Colorado	8,474,750	3,656,068	10,011,539	6,817,111	2,468,411	312.1	3,177,174	1,177,174	1,454,454	53.2		1,353,531	37.7	36,469	138,151	
Connecticut	2,865,740	1,494,168	2,011,380	2,310,712	221,251	48.0	1,253,194	795,028	920,127	23.2		134,513	1.3		377,277	
Delaware	1,419,068	923,395	2,207,097	1,494,448	576,536	110.8	433,426	223,925	200,447	18.5	578	376,683	7.8	167	146,013	
Florida	5,231,834	2,661,345	6,903,853	5,074,227	705,545	235.6	1,581,213	69,148	1,466,517	72.7	23,968	217,869	5.0	64,121	150,298	
Georgia	10,091,185	5,113,491	9,067,538	7,800,494	971,664	539.0	3,628,865	1,419,406	1,703,148	178.9	6,394	217,869	5.0	464,591	2,221,250	
Idaho	4,446,289	2,277,445	5,395,529	4,273,123	788,715	423.4	1,119,277	131,759	924,290	57.2		36,570	3.3	81,367	147,811	
Illinois	17,570,770	8,921,401	12,008,564	12,173,189	285,322	311.4	11,858,659	5,304,438	6,554,131	228.8	50,260	1,270,538	11.0	42,823	850,490	
Indiana	10,037,443	5,048,963	8,200,128	8,027,933	137,034	265.3	5,678,057	1,658,495	3,965,895	18.4	226,066	564,302	18.4	125,350	397,732	
Iowa	10,055,660	5,118,351	10,612,864	9,278,137	988,005	435.2	4,986,043	676,717	3,509,891	377.2		167,600	8.7	746	516,865	
Kansas	10,089,604	5,117,675	11,886,562	9,551,346	1,283,464	568.4	4,177,107	735,259	3,171,269	14.1	92,417	193,011	1.3	8,719	366,639	
Kentucky	7,517,359	3,818,311	7,659,329	6,571,359	593,581	248.4	3,695,127	1,177,174	1,711,869	24.6	30,000	193,011	1.3	106,125	366,639	
Louisiana	5,422,591	2,963,932	5,011,374	4,279,444	342,291	125.3	3,432,133	1,340,921	1,895,906	80.9	192,479	331,462	13.4	15,747	332,282	
Maine	3,369,917	1,711,565	4,182,421	3,256,284	756,964	173.5	892,090	34,506	784,708	18.2		154,253	2.4	19,170	11,661	
Maryland	3,564,527	1,810,598	2,280,231	2,040,690	183,820	90.8	2,461,446	1,179,136	1,488,730	35.5	9,800	222,910	6.3	334,901	914,478	
Massachusetts	5,597,100	3,350,174	4,200,967	3,620,889	98,399	57.6	4,604,125	2,929,293	1,688,898	31.0		137,670	4.6	17,152	1,249,387	
Michigan	12,537,337	6,446,646	12,537,337	11,269,174	3,268,163	423.2	6,755,794	1,370,637	5,385,207	247.9	19,400	447,225	40.7	116,420	217,938	
Minnesota	10,686,569	5,405,554	12,727,520	9,574,333	2,756,603	1,382.2	2,937,995	820,181	1,903,508	216.3	64,220	308,065	24.7	133,635	467,378	
Mississippi	6,978,675	3,540,227	8,078,765	4,964,423	820,235	473.0	3,887,419	1,809,872	1,462,473	208.1	60,531	776,653	49.3	143,788	460,866	
Missouri	12,180,306	6,173,740	11,550,962	10,077,170	689,581	943.8	5,682,107	1,694,996	3,817,694	490.1	232,230	1,666,261	60.1	232,230	1,666,261	
Montana	7,459,734	3,769,734	9,998,341	7,102,115	2,599,133	897.3	1,304,204	68,540	1,118,507	96.6	147,641	203,506	29.9	121,384	58,788	
Nebraska	7,422,961	3,964,154	10,124,912	7,352,697	1,261,400	853.6	2,795,165	21,644	2,349,925	159.7		112,535	11.0	44,620	198,902	
Nevada	4,495,917	2,302,386	5,569,922	4,283,523	1,208,893	532.7	1,083,897	216,818	869,430	175.3	52,327	224,137	18.0	23,249	62,479	
New Hampshire	1,909,839	969,462	2,594,103	1,809,280	449,896	62.7	504,101	29,000	468,508	14.5	13,044	224,137	18.0	58,515	51,094	
New Jersey	6,346,039	3,220,879	5,283,742	4,891,660	138,352	61.4	3,104,136	1,295,764	1,475,092	22.1		812,020	2.1	158,615	737,455	
New Mexico	5,782,935	2,941,700	7,478,122	5,665,578	1,584,756	672.9	1,085,970	36,931	1,090,748	45.9	12,653	183,963	1.1	177,794	108,532	
New York	22,350,101	11,357,921	24,159,758	19,340,340	1,475,479	402.1	15,212,881	2,514,359	9,050,332	413.3	74,732	238,090	7.7	405,720	644,261	
North Carolina	9,522,293	4,840,941	10,828,732	7,997,214	1,746,315	1,061.5	3,006,699	927,344	1,995,461	242.4	182,695	544,144	32.8	145,077	591,861	
North Dakota	5,804,446	2,918,867	5,986,758	5,095,278	444,581	1,607.2	3,985,427	399,115	463,333	243.2	331,561	651,058	191.7	92,493	1,369,296	
Ohio	15,484,532	7,865,012	17,011,942	15,326,171	896,629	594.5	6,284,657	370,463	5,395,716	165.4		107,800	6.5	87,958	1,514,871	
Oklahoma	2,216,758	4,695,180	9,785,336	8,414,467	977,677	693.1	4,905,897	792,103	2,910,930	135.4	6,016	286,985	14.0	4,272	379,289	
Oregon	18,591,004	9,599,788	19,869,391	18,069,391	2,155,200	777.7	8,259,384	1,777,923	6,117,936	282.2	2,126	430,493	7.9	165,564	829,715	
Pennsylvania	1,998,708	1,014,372	2,142,381	1,858,334	204,739	66.7	710,175	79,740	614,156	16.9		36,815	1.4	60,634	154,863	
Rhode Island	5,499,165	2,770,344	5,080,344	4,646,084	365,578	391.7	2,462,517	735,422	1,677,694	229.9	27,225	66,990	4.4	20,430	664,632	
South Carolina	6,011,479	3,047,545	6,206,510	5,066,355	668,338	1,103.5	2,100,513	678,459	1,372,984	361.5	40,496	362,719	40.4	246,258	643,602	
South Dakota	4,402,618	2,201,911	6,400,710	7,692,307	1,007,766	374.9	3,041,606	697,892	2,297,311	98.5	20,080	430,530	7.8	131,714	577,343	
Tennessee	24,344,044	12,291,793	26,570,946	22,965,926	2,411,912	2,168.4	9,244,000	994,287	7,907,309	520.1	52,140	1,111,107	66.7	271,688	427,010	
Texas	4,134,708	2,132,691	5,469,732	3,926,216	946,763	512.9	1,534,328	260,153	1,010,350	88.7		17,000	1.1	8,337	138,577	
Utah	1,867,573	944,007	2,467,174	1,806,908	387,121	116.6	573,768	26,401	510,740	22.1	4,042	43,110	3.9	29,421	7,236	
Vermont	7,416,757	3,765,287	8,352,716	6,644,620	1,090,593	493.4	2,804,093	558,166	1,971,201	149.2		507,740	27.9	52,416	197,864	
Washington	6,115,867	3,106,412	7,036,154	5,832,732	1,127,173	535.5	2,214,157	281,591	1,895,367	42.9	161,555	537,868	5.3	1,845	35,618	
West Virginia	4,474,234	2,280,235	4,249,906	3,738,906	387,201	146.5	1,655,705	671,325	984,340	49.5	56,511	142,561	7.9	7,492	766,132	
Wisconsin	9,724,881	4,941,637	10,718,068	9,092,904	1,184,252	460.6	3,937,931	332,263	3,179,750	133.1	82,664	400,632	7.6	57,290	169,097	
Wyoming	4,501,567	2,287,112	5,125,126	4,123,788	817,355	719.2	1,707,898	363,973	1,316,604	31.1		47,735	5.2	13,566	69,021	
District of Columbia	1,918,469	971,462	2,038,246	1,668,009	370,246	16.9	691,229	250,164	393,565	2.7	20,973	75,395	4.6	295	115,097	
Puerto Rico	1,471,562	949,718	1,007,538	704,442	104,442	84.2	1,430,937	1,126,895	93,424	20.6		345,645	6.0	14,733	504,709	
TOTALS	394,000,000	200,000,000	864,105,312	745,439,500	26,000,000	26,000,000	170,445,724	41,523,397	114,215,941	7,482.9	2,516,601	17,531,538	876.3	4,495,014	22,413,021	